# Table F19 THERMODYNAMICS FORMULAE

		·
Dennity (p) p = matr volume	<u>m</u>	Pirial Equations
Specific volume (v) $v = \frac{volume}{mass}$	$-\frac{V}{m} = \frac{1}{\rho}$	$P = RT/(\nu - B)$ $B_{\sigma} = BP_{\sigma}/RT_{\sigma}$
Specific weight (y) $y = \frac{\text{weight}}{\text{solves}}$		$z=1+(B_rP_r/T_r)$ $B^{(0)}=0.083-0.422/T_r^{1.6}$
Continuity equation $m = \sum_i \rho_i Y_i$	$t = \sum_{m=0}^{\infty} p E d$	$B^{(1)} = 0.139 - 0.172 / T_e^{4.2}$ $B_e = B^{(0)} + coB^{(1)}$
Ideal-gas Equation of State $p_V = \frac{\overline{R}}{M}$	. R = universal gas constant M = molar mass of substance	
Snergy balance for an open system	0-# = m. (v. + p.v. + p. v. +	(1+gs.) -m.(v. + p.v. + V1+gs.) (v.)
Nork of steady - flow, open system	$W_{ln} = \int v  dp + \Delta k e + \Delta p$	
thermal efficiency of heat engine	n - for - 0 - 0 -	$\eta_i = \frac{T_{H} - T_L}{T_{H}}$
Coefficient of performance (refrigerator)	COPs - Par	$COP_A = \frac{T_b}{T_B - T_A}$
Coefficient of performance (heat pump)	$COP_{MP} = \frac{Q_{mi,N}}{W_{co}}$	
Change in entropy	$\Delta S = \int \left(\frac{\delta Q}{T}\right)_{\frac{1}{T}} = S.$	2-5, (KJ/K)
Shange in entropy for open system	dS (***) + 1,6m,	- 1,000, + Syen (47/1 5; - mese + Syen (47/1 + 8 10 1/2; \(\sigma_{\color=0}^{\color} \color_{\sigma}^{\color} - 5'_1 - 5'_1
Change in entropy for ideal gas = dS =	e, # - 8 to F; ds = 1, e, #	$+ B \ln \frac{V_2}{V_f};$ $\int_{T_c}^{T_c} e_\rho \frac{dT}{T} - S^a_\rho - S^a_\rho$
sentropic Turbine Efficiency	ητ = work of actual expan work of reversible exp	ration = h <sub>J</sub> - h <sub>D</sub>
Idiahatic Compressor Efficiency	m <sub>c</sub> = work of severable comp work of actual compr	restion h <sub>i</sub> h <sub>i</sub>

Expression for a Reversible Process (Expansion/ Compression) in which PV' = constant

Isothermal		Adiabatic, c.pe, (7)	
au.	•	$\frac{nR}{\gamma - 1}(P_2 - P_1)$ or $\frac{1}{\gamma - 1}(P_2V_2 - P_1V_1)$	
Q	$-nRT \ln \left( \frac{P_0}{P_0} \right)$	o	
w	nRT le(D)	$\frac{nR}{\gamma-1}(P_2-P_1) \text{ or } \frac{1}{\gamma-1}(P_2V_2-P_1V_1)$	
Y	1	e, ke,	

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**James Alfred Ewing** 

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