

### CHEMISTRY Terminology The ABCs of chemistry

A: Symbol for atomic mass marrier

absolute zero: Lowest temperature; 0 Kelvin

Act Symbol for the element actinium, Z = 89.

accuracy: The closeness of measurement to true value.

acetate anion: CallaDy, charge -1

postic acid: C-01.O-

acetome: Directled ketome: organic solvent.

acid-base indicator: A chemical that changes color lift the pHI champes

acid-base titration: A method for determining acid or base concentrations.

noid definition: See Archenius acid, Brensted-

Lower sold, Lewis sold.

acid ionization constant (K.): Equilibrium constant for acid dissociation.

acidic solution; pit Buckeyer 1

actinide: Element switch 2: ~ 90 or allower: radiosective.

activation energy

(IE.)): A prescess is energy.

Resident

Е.

**Activation Energy** 

Product

adhesion: Attraction of unlike molecules or

Age: Symbol for the element silver, Z = 47

Alt: Symbol for the element aluminum, Z = 13.

alcohol: Organic compound with -OH group; BOH aldehyde: Carboxyl group bonded to hydrogen and an organic group

alkali metal family: Lithium, sodium, potosium, nubidium, opsium, and francium; column #1

alkaline earth metal family: Beryllium, magnesium, calcium, strontium, barium, and radium; column #2.

allkame: Hydrocurbon; all C-C single bonds

alkene: Hydrocarbon; I or more C+C double bonds. alloyse: Hydrocarbon, I ar more CFC triple bonds.

allotropes: Two or more forms of an element

alticy: Solution of 2 or more metals.

alpha (a): Greek letter that denotes radioactive particle and various scientific variables.

alpha particle: Helium micleus; charge +2

Amc Sverbol for the element sovericium, Z = 95

amalgame Alloy of mercury with other inetal

amine: Organic base; RRRN; modified amountain

amino acid: Compound with organic acid and organic

base property; forms proteins and peptides

ammonia: Nill, base

ammonium cation: NIL, charge +1.

ammonium hydroxide: NHLOH

amphoteric oxide: Exhibits both sold and basic

properties

named Assessing process until \$500 masses of C-12.

arrion; ton with a regutive charge

anode: Electrole that supports existation

antibonding MO: MO is less stable than separate AOs.

AO: Abbreviation for atomic orbital

augt Asparents "dissolved in water"

Art: Symbol for the element argon, Z = 18.

aromatic: Organic compound with a benome ring

Arrhenius acid: Produces hydronium ion in water

Arrhenius basec Yields hydroxide ion in water

Arrhenius equation for rate constant (k):

 $k = Ae^{-Ea/RT}$ 

R. - activation energy

arsenide anion: Av., charge «It

As: Symbol for the element assenic, Z = 33.

All: Symbol for the element astation, Z = 85

aten: Symbol for pressure in "atmospheres"

atom: Fundamental unit of all matter

atomic mass number (A): Total number of protons and neutrons.

atomic number (Z): Number of protom in the nucleus.

atomic orbital: Wave motion of electrons in atoms atomic radius: Impirical measure of atom size

atomic weight: Weighted average of natural isotopes of an observed

Auc Symbol for the element gold, Z = 79

Aufbau principle: Guides the filling of electronic and solve the self that educations to

Avogadro's law; Volume is proportional to gas moles at fixed pressure and temperature

Avogadro's number: N<sub>s</sub> = 6.022 × 10<sup>th</sup> met<sup>-1</sup>

B: Symbol for the element boron, Z = 5

Ba: Symbol for the element barium, Z = 56

balanced equation: Each side of the equation has the same number of atoms of each element and charge

balanced redox: Electron loss = electron gain

barium cation: Bar, charge +2.

base definition: Sor Arrhenius base, Breested-

Lovery base, Lowis base

base ionization constant (K.): Liquilibrium constant for home dissociation.

basic solution; plif over 7

battery: Source of electrical power; galvanic cell-

Bo: Sambol for the element beryllium, Z = 4benzene: C.H.: ereanic solvent. Benzene

aromatic ring of 6 carbon atoms beryllium eation: No., charge

both (80: Greek letter that denotes radioactive particle and a number of scientific variables

beta particle: Enemetic electron from the mucleon; charge -1.

Bh: Symbol for the element bohrium, Z = 107

Bit: Symbol for the element binemath, Z = 83blearbonate anion: HCOs, charge -1

bimolecular oxygen: O:

binary compound: Fermed from 2 elements

bisulfate anion: HSO/, charge -1

Bk: Symbol for the element berkelium, Z = 97

bleach: Chemical that is a strong exidizing agent

boiling: Liquid -- gas at the boiling point

boiling point elevation: A solution has a higher

bodling point than a paint solvent

boiling point (T<sub>s</sub>): Liquid-gas equilibrium, P = 1 atm

boiting point of water; 100°C

bond energy: Energy held by a chemical bond

bond length: Distance between 2 atoms in a bond

bond order for Lewis model: it of bonds divided by If of bounded atomics

bond order for MO treatment: # of filled bonding MOs minus # of filled antibonding MOs

bonding electrons: Form chanical bonds in a

bonding MO: MO is more stable than AOs

Biography barry PW

constant for gas, fixed T and term

By: Sumbot for the element becoming, 2 = 35

bromide anion: fit.

charges -1 Bronsted-Lowery acid:

Proton donor Bronsted-Lowry base: Proton acceptor

buffer: Composed of weak acid and weak base; serves

to keep pHI constant

cc Symbol for penti-, the SI porfex for V<sub>mi</sub> or 10.7

Biowte's Law

200 400 400 800

Carboxylie Acid

R-C-OH

et: Symbol for the speed of light.

\*C: Celsius temperature scale

C: Sconbol for the element curbon, Z = 6.

Culti/Ost Acettic acid

Cat Symbol for the element calcium, Z = 20

calcium cation: Ca\*, charge +2

exclusions theoriday Culti-

calcium hydroxide: CarOffle catories linglish unit of heat energy

calorimetry: Study of the release or absorption of heat.

carbide anion; C\*, charge -4

carbohydrate: Organic compound; carbon bonded to neveral -OH groups (sugar and stanch).

carbon dioxide: COcarbon monoxide: CO

carbonate anion: COV, charge -2.

carbonic acid: IE-CO.

carbonyl group: CO, ligand complex carboxyl group; >CO banded to 2 groups carboxyllic acid: Organic acid: R-COOH

catalyst: Accelerates a reaction but is not consumed in the reaction; lowers the activation

COUNTY OF

cathode: Electrode that supports reduction.

eathers less with a positive charge

Cd: Symbol for the element cadmium, Z = 48.

# **Thermal Chemistry Study Guide**

**Zhe Xu** 

#### **Thermal Chemistry Study Guide:**

Chemistry, Study Guide James E. Brady, John R. Holum, 1993-02-11 Offers accurate lucid and interesting explanations of basic concepts and facts of chemistry while helping students develop skills in analytical thinking and problem solving Students are taught in a variety of ways to think of skills as tools that can be used to solve complex problems Several aids are included to help focus and inspire student interest frequent reference to common chemicals in commercial products numerous photographs of reactions in chapter practice exercises following worked examples Thermodynamics Problem Solving in Physical Chemistry Kathleen E. Murphy, 2020-03-23 Thermodynamics Problem Solving in Physical Chemistry Study Guide and Map is an innovative and unique workbook that guides physical chemistry students through the decision making process to assess a problem situation create appropriate solutions and gain confidence through practice solving physical chemistry problems The workbook includes six major sections with 20 30 solved problems in each section that span from easy single objective questions to difficult multistep analysis problems Each section of the workbook contains key points that highlight major features of the topic to remind students of what they need to apply to solve problems in the topic area Key Features Provides instructor access to a visual map depicting how all equations used in thermodynamics are connected and how they are derived from the three major energy laws Acts as a guide in deriving the correct solution to a problem Illustrates the questions students should ask themselves about the critical features of the concepts to solve problems in physical chemistry Can be used as a stand alone product for review of Thermodynamics questions for major tests Study Guide to Organic Chemistry Robert Thornton Morrison, Robert Neilson Boyd, 1975 An Elementary manual of chemistry Francis Humphreys Storer, 1894 Basic Inorganic Chemistry F. Albert Cotton, Geoffrey Wilkinson, Paul L. Gaus, 1995-01-18 Explains the basics of inorganic chemistry with a primary emphasis on facts then uses the student's growing factual knowledge as a foundation for discussing the important principles of periodicity in structure bonding and reactivity New to this updated edition improved treatment of atomic orbitals and properties such as electronegativity novel approaches to the depiction of ionic structures nomenclature for transition metal compounds quantitative approaches to acid base chemistry Wade s rules for boranes and carboranes the chemistry of major new classes of substances including fullerenes and silenes plus a chapter on the inorganic solid state An Elementary Manual of Chemistry William Ripley Nichols, William B. Lindsay, Frank Humphreys Storer, 1894 General Catalogue of the Books Except Fiction, French, and German, in the Public Library of Detroit, Mich Detroit Public Library, 1889 The Best Books: a Readers Guide to the Choice of the Best Available Books (about 50.000)... William Swan Sonnenschein, 1895 **Advances and Trends in Engineering** Sciences and Technologies II Mohamad Ali, Peter Platko, 2016-11-30 These are the proceedings of the 2nd International Conference on Engineering Sciences and Technologies ESaT 2016 held from 29th of June until the 1st of July 2016 in the scenic High Tatras Mountains Tatransk Matliare Slovak Republic After the successful implementation and excellent feedback

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