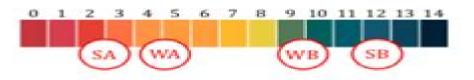
## Acid-Base Solutions - PhET Simulator

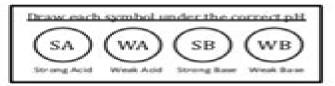
Instructions: Open the Acid-Base Solutions simulator on the PhET website/app. Choose "Introduction."

- > pH Test Strips change color when they come in contact with a solution.
- The color of the strip is compared to a pH Color Key to determine the pH.

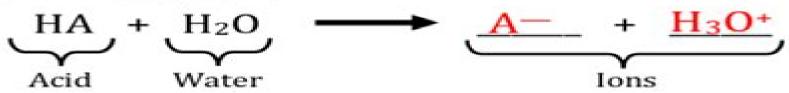


- Select the pH Test Strip from the Tools menu. Dip it into water. What is the pH?
- 2) Use pH test strips to test the <u>Strong Acid</u>, <u>Weak Acid</u>, <u>Strong Base</u>, and <u>Weak Base</u>.
  Draw the symbol for each solution in the appropriate location on the pH Color Key below:





- When acids and bases dissolve in water, their ions dissociate (separate).
- ➤ Acids dissociate by releasing H\* ions. These H\* ions bond to H2O to form H3O\*.
- 3) Choose the Strong Acid. Complete the chemical formula for dissociation of a strong acid:



Instructions: Under "Views," choose the Graph option.

- 4) What is the concentration of Ions (A and H3O) in the Strong Acid? \_\_
- 5) What is the concentration of <u>lons</u> (A and H<sub>3</sub>O\*) in the Weak Acid?
- 6) Which type of acid (Strong or Weak) produces more ions?

- 3.16 x 10<sup>-5</sup>
- Strong Acid

# **Worksheet 101 Acid Base And Ph**

**Alicia Grosso** 

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