



Acid, Bases And Salts

- These are the substances which have sour taste.
- They turn blue litmus solution red.
- They give H^+ ions in aqueous solution.
- The term 'acid' has been derived from the Latin word, acidus, which means sour.

Strong Acids : HCl , H_2SO_4 , HNO_3

Weak Acids : CH_3COOH , Oxalic acid, Lactic acid

Concentrated Acid : Having more amount of acid + less amount of water

Dilute Acid : Having more amount of water + less amount of acid

- These are the substances which are bitter in taste and soapy in touch.
- They turn red litmus solution blue.
- They give OH^- ions in aqueous solution.

Strong Bases : NaOH , KOH , Ca(OH)_2 .

Weak Bases : NH_4OH

Alkalis : These are bases which are soluble in water [NaOH, KOH, Ca(OH)₂].

These are the compounds formed from reaction of acid and base.

Topic 10 Acids Bases And Salts Practice Questions

The Princeton Review



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