

PROTOKOLL

Name: Ana Lucía Ordoñez López

Thema:

Materialien: Scharfes Küchenmesser, 3 mittelgrosse Äpfel, 12 Bechergläser 400 mL, Haushaltzucker (Saccharose), destilliertes Wasser, Stoppuhr, digitale Waage, Küchenpapier und Lineal.

Entwurf: Bestimmung des Zuckergehalts in den Zellen von Apfelstücken mittels Messung ihrer Längenveränderung in Lösungen unterschiedlicher Zuckerkonzentration.

Definition des problems: Apfelstücke, die ungefähr gleich schwer sind, werden in verschiedene Lösungen mit unterschiedlichen Zuckergehalten gegeben und nach 30 Minuten wird das Gewicht noch mal gemessen.

Grundlagen: Im Zytoplasma von Pflanzenzellen sind unterschiedliche Substanzen wie Zucker und Salze gelöst. Diese sind an den biochemischen Prozessen des Zellstoffwechsels beteiligt.

Bei einer Plasmolyse schrumpft der gesamte plasmatische Inhalt einer Pflanzenzelle, wobei sich die Plasmamembran von der Zellwand ablöst. Dieses passiert, wenn sich die Zelle in einem hypertonen Medium befindet. Dann gibt die Zelle osmotisch Wasser in das umgebende, höher konzentrierte Medium ab und der Protoplast wird kleiner. In einem hypotonischen Medium nimmt die Zelle osmotisch Wasser auf und wird grösser.

Den Zustand bei dem sich der Protoplast gerade beginnt von der Zellwand abzulösen, nennt man Grenzplasmolyse: Die Zelle ist nicht mehr voll turgeszent, aber auch noch nicht richtig plasmolysiert. Dies passiert in einem isotonischen Medium (die gleiche Konzentration an Teilchen wie im Plasma der Zelle). Damit kann man die Konzentration des Zytoplasmas indirekt bestimmen.

Hypothese:

1. Bei einem höheren Zuckergehalt der Lösung (=hypertonisch) als dem des Zytoplasmas in den Apfelzellen werden die Stücke leichter werden, da sie osmotisch Wasser abgeben.
2. Bei einem niedrigeren Zuckergehalt der Lösung (=hypotonisch) als dem des Zytoplasmas in den Apfelzellen werden die Stücke schwerer werden, da sie osmotisch Wasser aufnehmen.
3. Wenn der Zuckergehalt der Lösung genauso wie das Zytoplasma in den Apfelzellen ist, wird das Gewicht nicht verändert.

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Christian G. Meyer



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1. Understanding the eBook Wie Man Ein Kautabakpriemen Herstellt Ebook
 - The Rise of Digital Reading Wie Man Ein Kautabakpriemen Herstellt Ebook
 - Advantages of eBooks Over Traditional Books
2. Identifying Wie Man Ein Kautabakpriemen Herstellt Ebook
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Wie Man Ein Kautabakpriemen Herstellt Ebook
 - User-Friendly Interface
4. Exploring eBook Recommendations from Wie Man Ein Kautabakpriemen Herstellt Ebook
 - Personalized Recommendations
 - Wie Man Ein Kautabakpriemen Herstellt Ebook User Reviews and Ratings
 - Wie Man Ein Kautabakpriemen Herstellt Ebook and Bestseller Lists
5. Accessing Wie Man Ein Kautabakpriemen Herstellt Ebook Free and Paid eBooks
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 - Wie Man Ein Kautabakpriemen Herstellt Ebook eBook Subscription Services
 - Wie Man Ein Kautabakpriemen Herstellt Ebook Budget-Friendly Options
6. Navigating Wie Man Ein Kautabakpriemen Herstellt Ebook eBook Formats

- ePub, PDF, MOBI, and More
- Wie Man Ein Kautabakpriemen Herstellt Ebook Compatibility with Devices
- Wie Man Ein Kautabakpriemen Herstellt Ebook Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Wie Man Ein Kautabakpriemen Herstellt Ebook
 - Highlighting and Note-Taking Wie Man Ein Kautabakpriemen Herstellt Ebook
 - Interactive Elements Wie Man Ein Kautabakpriemen Herstellt Ebook
- 8. Staying Engaged with Wie Man Ein Kautabakpriemen Herstellt Ebook
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Wie Man Ein Kautabakpriemen Herstellt Ebook
- 9. Balancing eBooks and Physical Books Wie Man Ein Kautabakpriemen Herstellt Ebook
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Wie Man Ein Kautabakpriemen Herstellt Ebook
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Wie Man Ein Kautabakpriemen Herstellt Ebook
 - Setting Reading Goals Wie Man Ein Kautabakpriemen Herstellt Ebook
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Wie Man Ein Kautabakpriemen Herstellt Ebook
 - Fact-Checking eBook Content of Wie Man Ein Kautabakpriemen Herstellt Ebook
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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