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web gas stoichiometry practice problems using the following equation $2\text{NH}_3(\text{g}) + \text{H}_2\text{SO}_4(\text{aq}) \rightarrow (\text{NH}_4)_2\text{SO}_4(\text{aq})$ calculate the volume of NH_3 in liters needed at 20°C and 25 atm to react with 150 kg of H_2SO_4

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web gas stoichiometry worksheet please answer the following on separate paper using proper units and showing all work please note that these problems require a balanced chemical equation 1 carbon monoxide reacts with oxygen to produce carbon dioxide if 1.0 L of carbon monoxide reacts with oxygen at STP

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web balancing equations and simple stoichiometry key balance the following equations $1. \text{N}_2 + 3\text{F}_2 \rightarrow 2\text{NF}_3$ $2. \text{C}_6\text{H}_{10}\text{O}_2 + 12\text{O}_2 \rightarrow 6\text{CO}_2 + 10\text{H}_2\text{O}$ key solutions for the

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