

Unit 6: Molecules & Compounds

Key Terms

Ionic Bonding

Electron dot diagram
ion
anion
cation
chemical bond
ionic bond
chemical formula
crystals

Covalent Bonding

Covalent bond
molecule
polar covalent bond

Naming Compounds & Writing Formulas

Polyatomic ion

Essential Questions

Electron Configurations: **electron dot diagram** – a model of an atom in which each dot represents a valence electron.

PERIODIC TABLE ELEMENTS 1–20							
H ·							He ·
Li ·	Be ·	B ·	C ·	N ·	O ·	F ·	Ne ·
Na ·	Mg ·	Al ·	Si ·	P ·	S ·	Cl ·	Ar ·
K ·	Ca ·						

• What is a chemical bond?

A **chemical bond** is the force that holds atoms and ions together as a unit.

• How is an ionic bond different from a covalent bond?

Ionic Bonds: some elements achieve stable electron configurations through the transfer of electrons between atoms.



Formation of ions – when an atom gains or losses an electron, the number of protons is no longer equal to the number of electrons. The charge is no longer neutral.

ion – an atom with a net positive or negative charge. Represented by a + or – sign.

Cl gains an electron 17 protons 18 electrons charge of 1- Cl^{1-} or Cl^-

anion – (an eye un) an ion with a negative charge

Naming: using part of the element name plus the suffix *-ide*.

Cl^- : *Chloride* ion.

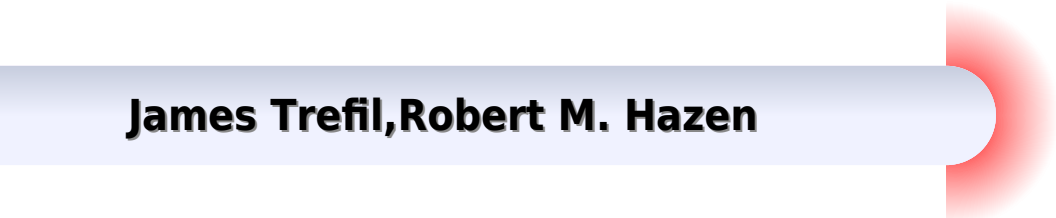
Na loses an electron 11 protons 10 electrons charge of 1+ Na^{1+} or Na^+

cation – (kat eye un) an ion with a positive charge

Naming: just use the elements name.

Unit 6 Study Guide Covalent Bonding

James Trefil, Robert M. Hazen



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Life: The Science of Biology Study Guide William K. Purves, Edward Dzialowski, Lindsay Goodloe, Betty McGuire, Nancy Guild, Paula Mabee, 2003-12-26 New edition of a text presenting underlying concepts and showing their relevance to medical agricultural and environmental issues Seven chapters discuss the cell information and heredity evolutionary process the evolution of diversity the biology of flowering plants and of animals and ecology and biogeography Topics are linked by themes such as evolution the experimental foundations of knowledge the flow of energy in the living world the application and influence of molecular techniques and human health considerations Includes a CD ROM which covers some of the subject matter and introduces and illustrates 1 700 plus key terms and concepts Annotation copyrighted by Book News Inc Portland OR

Student Study Guide for Biology [by] Campbell/Reece Martha R. Taylor, 2002 Marty Taylor Cornell University Provides a concept map of each chapter chapter summaries a variety of interactive questions and chapter tests

Student Guide For Living Chemistry Barbara Ucko, 2012-12-02 Student Guide for Living Chemistry is a 23 chapter textbook guide that allows students to study and review on their own and test their understanding to help them prepare for examinations Every chapter begins with a list of objectives stating exactly the skills to develop in a particular unit Each objective corresponds to a section in the textbook Living Chemistry Three kinds of questions are provided for each objective to check the student s understanding namely short answer Study Questions multiple choice and fill in The answers for all questions are provided at the end of the chapter The opening chapters cover the SI units composition of matter chemical bonding compounds chemical change gases respiration and water The subsequent chapters deal with solutions acids bases salts nuclear and organic chemistry oxygen derivatives and hydrocarbons polymers and other organic derivatives This textbook also explores the chemistry of carbohydrates lipids proteins enzymes and energy and carbohydrate metabolism The remaining chapters discuss the chemistry of vitamins hormones body fluid drugs and poisons Undergraduate chemistry students will find this book invaluable

Study Guide to Accompany Basics for Chemistry Martha Mackin, 1982-01-01 Study Guide to Accompany Basics for Chemistry is an 18 chapter text designed to be used with Basics for Chemistry textbook Each chapter contains Overview Topical Outline Skills and Common Mistakes which are all keyed to the textbook for easy cross reference The Overview section summarizes the content of the chapter and includes a comprehensive listing of terms a summary of general concepts and a list of numerical exercises while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as

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Study Guide for The Anatomy and Physiology Learning System Edith Applegate, 2010-03-30 Designed to accompany The Anatomy and Physiology Learning System 4th Edition by Edith Applegate this study guide helps you learn and review basic A P concepts Each chapter emphasizes medical terminology with a set of key terms word parts clinical terms and abbreviations and then adds a variety of fun filled learning exercises review questions a quiz and a word puzzle The study guide corresponds to the textbook chapter for chapter Chapter learning objectives help you focus on the most important material Key concepts are defined on the first page of each chapter in the workbook Learning exercises for each chapter include short answer matching and diagrams to label and color Self quizzes allow you to measure your progress and understanding Fun and Games features end each chapter with a variety of engaging puzzles covering words and concepts A chapter summary provides a brief review of each chapter A chapter review provides questions for reinforcement and review of the concepts in each chapter *Student Study Guide for Biology [by] Campbell/Reece/Mitchell* Martha R. Taylor, 1999

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