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Balancing Equations Worksheet Chemistry Project

2. 
$$4 - Ag1 + 1 - Fe_2(CO_3)_3 \rightarrow 2 - Fel_3 + 3 - Ag_2CO_3$$

3. 
$$1_{C_2H_4O_2} + 2_{O_2} \rightarrow 2_{CO_2} + 2_{H_2O}$$

5. 
$$_{_{_{_{_{2}}}}}$$
  $_{_{_{_{2}}}}$   $_{_{_{2}}}$   $_{_{_{5}}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$   $_{_{5}}$ 

8. 
$$3_{B_2}H_2SO_4 + 2_{B_3}B(OH)_3 \rightarrow B_2(SO_4)_3 + B_2 + B$$

9. 
$$1_{S_0} + 5_{O_2} \rightarrow 8_{SO_2}$$

# **Unit 7 Balancing Chemical Reactions Practice Test**

Robert C. Smoot, Smoot, Richard G. Smith, Jack Price

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