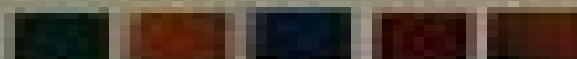


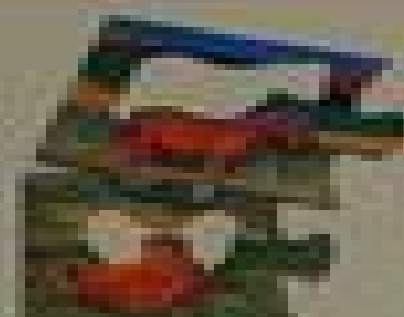


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web answer the mass percentage of an element of a solution is defined as the mass of the solute in grammes present in 100 g of the solution it s written like this
$$\frac{\text{mass of a component}}{\text{mass of component in solution total mass of solution}} \times 100$$
 end array

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web aug 13 2017 molarity practice problems this chemistry video tutorial explains how to solve common molarity problems it discusses how to calculate the concentration of a solution given the mass in grams