CAPE CHEMISTRY
UNIT 1: CHEMICAL PRINCIPLES AND
APPLICATIONS I
MODULE 1: FUNDAMENTALS IN CHEMISTRY

FORCES OF ATTRACTION VSEPR THEORY

MR. DANIEL C. WALSH, BBMS HEAD OF SCIENCE DEPARTMENT DECARTERET COLLEGE

VALENCE SHELL ELECTRON PAIR REPULSION THEORY

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- There is no direct relationship between the formula of a compound and the shape of its molecules. The shapes of these molecules can be predicted using a model developed about 30 years ago, known as the valence-shell electron-pair repulsion (VSEPR) theory.
- The VSEPR theory assumes that each atom in a

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