

This print-out should have 21 questions. Multiple-choice questions may continue on the next column or page – find all choices before answering.

001 10.0 points

An object undergoes a temperature change of $\Delta T = 10^\circ\text{C}$.

What would be another way of expressing the same change?

1. A $\Delta T = 10^\circ\text{C}$ is equivalent to a $\Delta T = 10^\circ\text{F}$.

2. A $\Delta T = 10^\circ\text{C}$ is equivalent to a $\Delta T = 10\text{K}$.

3. A $\Delta T = 10^\circ\text{C}$ is equivalent to a $\Delta T = 50^\circ\text{F}$.

4. A $\Delta T = 10^\circ\text{C}$ is equivalent to a $\Delta T = 283.15\text{K}$.

002 10.0 points

Convert 610°C to Fahrenheit.

1. 2030

2. 2660

3. 3650

4. 2300

5. 1130

6. 680

7. 1670

8. -58

9. 3200

10. 3830

Answer in units of $^\circ\text{F}$.

$$\begin{aligned} T_F &= 1.8T_C + 32 \\ &= 1.8(610) + 32 \\ T_F &= 1130^\circ\text{F} \end{aligned}$$

003 10.0 points

A building made with a steel structure is 610 m high on a winter day when the temperature is 0°F .

How much taller is the building when the temperature is 106°F ? The linear expansion coefficient of steel is $1.1 \times 10^{-5} (^\circ\text{C})^{-1}$.

1. 39.5144

2. 11.8983

3. 16.06

4. 35.5636

5. 52.1803

6. 29.161

7. 50.9826

8. 17.6287

9. 12.2564

10. 17.05

$$\begin{aligned} \Delta T &= 106 - 0 = 106^\circ\text{F} \\ \Delta T_C &= \frac{106}{1.8} = 58.9^\circ\text{C} \\ \uparrow \text{length} &= (610\text{ m})(1.1 \times 10^{-5})(58.9) \\ &= 0.395\text{ m} \\ &= 39.5149\text{ cm} \end{aligned}$$

$$\Delta L = \alpha L \Delta T_C \leftarrow \text{equation}$$

Answer in units of cm.

004 10.0 points

Under constant pressure and with a constant amount of gas present, what happens to the volume of the gas if the temperature increases constant V

1. Increases at first, then decreases

2. Increases

$$\frac{PV}{T} = \text{constant}$$

3. Decreases

4. Remains the same

5. Decreases at first, then increases

005 10.0 points

What is the volume of 2.6 mol of an ideal gas at a pressure of 3 atm and a temperature of 0°C ? 1 liter = 0.001 m^3 and 1 atm = 101300 Pascals.

1. 14.9373

2. 28.3809

3. 22.4059

4. 24.6465

5. 15.6842

6. 35.8495

7. 27.634

8. 33.6089

9. 17.9248

10. 19.4185

Answer in units of L

$$\begin{aligned} PV &= nRT \\ V &= \frac{nRT}{P} = \frac{(2.6)(8.314)(273\text{K})}{303900\text{ Pa}} \\ V &= 19.4185\text{ L} \end{aligned}$$

$$P = \frac{3\text{ atm}}{1\text{ atm}} \times \frac{101300\text{ Pa}}{1\text{ atm}} = \frac{303900\text{ Pa}}{10.13} = 30390\text{ Pa}$$

006 10.0 points

The absolute temperature of a sample of monatomic ideal gas is doubled at constant volume.

What effect, if any, does this have on the pressure and density of the sample of gas?

Pressure

Density

1. Remains the same

Remains the same

2. Doubles

Remains the same

3. Remains the same

Doubles

$$\begin{aligned} PV &= nRT \\ P &= \frac{nR}{V} T \end{aligned}$$

Ut Quest Physics Solutions

**Deog Ki Hong, Chang-hwan Lee, Hyun
Kyu Lee, Dong-pil Min, Tae-sun
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