- Which quantities must be conserved in all chemical reactions?
 - A) mass, charge, density
 - (B) mass, charge, energy
 - C) charge, volume, density
 - D) charge, volume, energy
- 2 Given the balanced equation:

I + I → I2 Which statement describes the process represented by this equation?

- A) A bond is formed as energy is absorbed.
- (B) A bond is formed and energy is released.
- C) A bond is broken as energy is absorbed.
- D) A bond is broken and energy is released.
- Given the formula for hydrazine:

How many pairs of electrons are shared between the two nitrogen atoms?

- B) 2
- C) 3
- D) 4
- What is the number of pairs of electrons shared in a. molecule of No?
 - AT I
- B) 2
- D) 6
- 5. Which atom in the ground state has a stable valence electron configuration?
- B) Al
- C) Si
- D) Na
- An atom of which element has the strongest attraction for electrons in a chemical bond?
- A chlorine

- B) carbon
- C) phosphorus
- D) sulfur
- 7. Which bond is least polar?
 - A) As-Cl

B) Bi-Cl

C) P-C1

Given the electron dot diagram:

H:F:

The electrons in the bond between hydrogen and fluorine are more strongly attracted to the atom of

- A) hydrogen, which has the higher electronegativity
- (B) fluorine, which has the higher electronegativity
- C) hydrogen, which has the lower electronegativity
- fluorine, which has the lower electronegativity
- Given the reaction:

$M + 2 \text{ H2O} \rightarrow M(\text{OH})_2 + \text{H2}$

The metal represented by M is most likely a metal. from Group

- A) I
- C) 11
- D) 13
- Which type of substance can conduct electricity in the liquid phase but not in the solid phase?
 - (A) ionic compound
 - B) molecular compound
 - C) metallic element
 - D) nonmetallic element
- The chemical bonding in sodium phosphate, NasPO₄ , is classified as
 - A) ionic, only
 - B) metallic, only
 - Doth covalent and ionic
 - D) both covalent and metallic
- 12. Which type of bonding is found in all molecular substances?
 - (A) covalent bonding
- B) hydrogen bonding
 - C) ionic bonding
- D) metallic bonding
- Which type of substance is soft, has a low melting. point, and is a poor conductor of heat and electricity?
 - A) network solid
- - molecular solid D) ionic solid
- C) metallic solid

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Marquis Who's Who

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web sep 15 2011 recently browsing hi everyone this is my first posting and like most newcomers i have been watching the site for some time my question relates to shotshell reloading and is probably aimed at the older members i have just acquired some nobel 60 reloading powder and i have been unable to find any loading data

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web apr 19 2015 posts 138 victoria top re nobel 60 powder by oldbloke sun apr 19 2015 10 39 am yes definitely shot gun powder i used it up until 20 or more yrs ago i have the 2000 adi manual not in there if i recall it was a middle of the road 12g powder i found this in an old reloading simplified book

load data for nobel powders shotgun forum - Jul 14 2023

web jun 4 2011 your nobel 0 powder is a rifle powder 30 06 etc and there is no shotshell data for it and the nobel 78 is a shotshell powder but the only cases it gives loads for are parallel tube types using fibre wads or eley monowads in 28 or 30 gram and eley primers the pressures vary from 2 35 to 2 65 ton s per square inch