

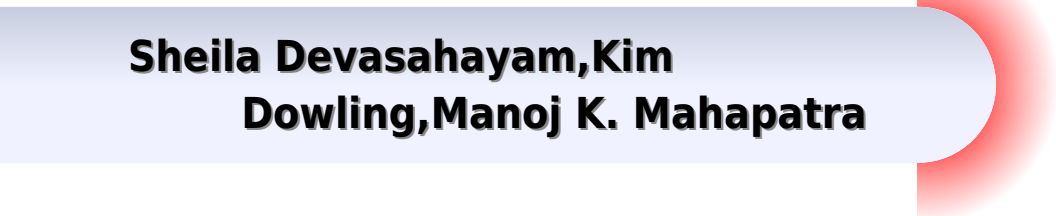
THE UNIVERSITY OF ZAMBIA 2024 INTAKE (UNDERGRADUATE FULL -TIME)SUCCESSFUL APPLICANTS

Student Id	Surname	First Name	Program Description	School Description
2023000009	MOTO	KITANO	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023000010	MYIYALAMBA	HEKAMU	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023000014	MSEWE	ANTHONY	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023000020	MAYUMBELO	JOY	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023000046	MUTALE	SAVIN	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023000047	KAPUNDA	STEVEN	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023000111	MUKULUDU	MUKUNGU	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023000112	MUSAMBA	EDMONS	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023000200	NYAMBE	THABO	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023000305	CHUNGU	JANET	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023000351	ZHABA	ABUTY	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023000479	BANDA	MUNA	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023000488	NG'ANDWE	ROSEMARY	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023000491	MUNGA	EDNA	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023000492	MWINGA	RONALD	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023000499	BANDA	DONORO	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023000477	CHIMVUSO	OLIVE	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023000591	MUSALAMBA	INDIGE	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023000696	MUSENGE	GRYEN	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023007585	ISIKETI	ISIKETI	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023007674	CHUNGU	EDGAR	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023000970	MUCHIMBA	MUCHIMBA	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023010063	SARALA	NYUMASE DICKSON	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023010075	SHUNGWIRE	NOAH	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023011853	NYIRENDA	TAGISA	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023011892	KACHEHA	OMANDA	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023011481	SQUIRE	DOMFORT	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023011524	LIEMBA	LIEMBA	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES

2023014514	MURANDO	DROOLWE	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023014971	KALUNGA	SPACE	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023015171	KWINGA	FRIDA	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023015484	NYIRENDA	ALEX	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023016649	ZULU	YAMKO	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023017009	AKETATA	DORCAS	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023017033	TEMBO	NELSON	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023017035	MWINGA	MWENYA	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023017623	MWENYWA	MWENYU	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023018749	CHUNGU	FRASE	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023018833	PHYBI	DAVID	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023018938	NGOMBA	KONDIVELANI	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023019486	MUKONOLA	SIBONGILE	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023019950	CHUNGA	MAZI	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023020433	MWANDAWIRE	WREZZY	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023020641	MATAMBA	MAYOYA	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023020760	CHINDU	BRENDA	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023020887	MUNYANGA	REBEKAH	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023021782	MUSHALA	LUNGWE	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023022489	SHANYA	GIFT	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023022711	KASUNGU	MARIA	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023022967	KOBYU	EMMANUEL	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023023114	MUDONGA	MELLEN	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023023572	KASONGO	ABISAIL	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023024200	KALUNGA	FORTUNE	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023025133	NYUMBE	COLLEKE	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023025575	KOBYU	DENNY	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023025885	MWANGA	PATRICIA	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023027004	LIPASHI	GRISON	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES
2023027691	MWANGA	WREZI	BACHELOR OF AGRICULTURAL SCIENCES	AGRICULTURAL SCIENCES

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**Sheila Devasahayam, Kim
Dowling, Manoj K. Mahapatra**



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YouthXchange training kit on responsible consumption for Africa Martinez-Roca, Carme,Thoresen, Victoria,UNEP,2017-12-31 **Sustainability in the Mineral and Energy Sectors** Sheila Devasahayam,Kim Dowling,Manoj K. Mahapatra,2016-09-15 Sustainable practices within the mining and energy sectors are assuming greater significance due to uncertainty and change within the global economy and safety security and health concerns This book examines sustainability issues facing the mining and energy sectors by addressing six major themes Mining and Mineral Processing Metallurgy and Recycling Environment Energy Socioeconomic and Regulatory and Sustainable Materials and Fleets Emphasizing an integrated transdisciplinary approach it deliberates on optimizing mining productivity and energy efficiency and discusses integrated waste management practices It discusses risk management cost cutting and integration of sustainable practices for long term business value It gives a comprehensive outlook for sustainable mineral futures from academic and industry perspectives covering mine to mill optimization waste risk and water management improved efficiencies in mining tools and equipment and performance indicators for sustainable developments It covers how innovation and research underpin management of natural resources including sustainable carbon management Focuses on mining and mineral processing metallurgy and recycling the environment energy socioeconomic and regulatory issues and sustainable materials and fleets Describes metallurgy and recycling and uses economic environmental and social parameter analyses to identify areas for improvement in iron steel aluminium lead zinc copper and gold production Discusses current research on mining performance indicators for sustainable development sustainability in mining equipment risk and safety management and renewable energy resources Covers alternative and conventional energy sources for the mineral sector as well water treatment and remediation and energy sustainability in mining Provides an overview of sustainable carbon management Offers an interdisciplinary approach with international focus **The African Film Industry** UNESCO,2021-10-01 The production and distribution of film and audiovisual works is one of the most dynamic growth sectors in the world Thanks to digital technologies production has been growing rapidly in Africa in recent years For the first time a complete mapping of the film and audiovisual industry in 54 States of the African continent is available including quantitative and qualitative data and an analysis of their strengths and weaknesses at the continental and regional levels The report proposes strategic recommendations for the development of the film and audiovisual sectors in Africa and invites policymakers professional organizations firms filmmakers and artists to implement them in a concerted manner The Emergence of Teacher Education in Zambia Brendan P. Carmody,2020-05-01 This book offers a detailed history of the development of teacher education in Zambia Also analysed is the nature of education offered at different times and how the teacher and his her education reflect this arguing the need for a fundamentally new philosophy of education and a mode of teacher formation in line with it *Integrated Approaches to Health: Concepts and Experiences in Framing, Integration and Evaluation of One*

Health and EcoHealth Simon R. Rüegg, Sandra C. Buttigieg, Flavie L. Goutard, Aurélie Binot, Serge Morand, Séverine Thys, Hans Keune, 2019-10-21 Integrated approaches to health address health challenges arising from the intertwined spheres of humans animals and ecosystems This eBook is the product of an interdisciplinary effort to establish how One Health EcoHealth and other integrated approaches to health are conceptualized framed implemented and evaluated today It supplements the handbook for the evaluation of One Health published by the COST Action Network for Evaluation of One Health NEOH with in depth reflections on the theory behind integrated approaches to health and One Health more specifically a brief version of the NEOH evaluation framework a supplementary evaluation approach and eight case studies in which the NEOH framework was applied The eBook is intended for practitioners researchers evaluators as well as funders of integrated approaches to health and beyond Without the outstanding support and leadership from the management committee this work would not have been achieved Our gratitude goes to Maria Eleni Filippitzi BE Vronique Renault BE Nihad Fejzic BA Sabina Seric Haracic BA Nenad Turk HR Relia Beck HR Luca Guardabassi DK Liza Rosenbaum Nielsen DK Flavie Goutard FR Vladimir Grosbois FR Brigitte Petersen DE Martin Hamer DE Elias Papadopoulos GR Ilias Chaligiannis GR Gabor Fldvri HU Anthony Staines IE Helen O Shea IE Shimon Harrus IL Gad Baneth IL Valeria Grieco IT Maurizio Aragrande vice chair IT Jovita Maikien LT Sandra Buttigieg MT Elaine Lautier MT Helmut Saatkamp NL Kitty Maassen NL Vlatko Ileski MK Mijalce Santa MK Merete Hofshagen NO Yngvild Wasteson NO Paulo Roriz PT Jorge Torgal PT Andrei D Mihalca RO Razvan Chereches RO Dragan Mili evi RS Sara Savic RS Joze Staric SI Mojca Juri i SI Pedro Soto Acosta ES Francisco Gimenez Sanchez ES Ann Lindberg SE Josef J rhult SE Jakob Zinsstag CH Simon Rüegg CH Barbara Hsler chair UK K Marie McIntyre UK Martha Betson UK Marieta Braks NL Chinwe Ifejika Speranza DE Spela Sinigoj SI Martijn Bouwknegt NL Andras Lakos HU and their substitutes Merel Postma BE Semra Cavajuga BA Estella Prukner Radovic HR Maria Vang Johansen DK Elena Boriani DK Ricarda Schmithausen DE Maryla Hanna Obszarski DE Smaragda Sotiraki GR Theofilos Papadopoulos GR Barry McMahon IE Massimo Canali IT Fabrizio Cecilian i IT Daniele De Meneghi IT Dalia Jurevi it LT Miroslav Radeski MK Toni Vekov MK Manuela Vilhena PT Carla Maia PT Alexandru Coman RO Branka Vidic RS Gospava Lazi RS Ksenija Sinigoj Gacnik SI Juan Gabriel Cegarra Navarro ES Asta Tvarijonaviciute ES Jos Cer n ES Helene Wahlstr m SE Karin Artursson SE Laura Cornelsen UK Jonathan Rushton UK We also would like to thank the 240 researchers that have engaged with the COST Action throughout and participated actively Our gratitude also goes to the Royal Veterinary College in London who acted as a grant holder Acknowledgments This publication is based upon work from COST Action TD1404 Network for Evaluation of One Health NEOH supported by COST European Cooperation in Science and Technology COST European Cooperation in Science and Technology is a funding agency for research and innovation networks Our Actions help connect research initiatives across Europe and enable scientists to grow their ideas by sharing them with their peers This boosts their research career and innovation www.cost.eu *Cancer Metabolism: Current Knowledge and Perspectives*

Leonardo Freire-de-Lima, Lucia Mendonça-Previato, Luciana Boffoni Gentile, 2019-10-23
Inclusive Growth in Zambia Pamela Kasese Bwalya, 2011

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web the purpose of the copper iron stoichiometry lab was to determine which of the two possible iron sulfate compounds formed when reacting aqueous copper ii sulfate with solid iron metal the techniques used in this lab were weighing by difference quantitative transfer and vacuum filtration

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web objective to apply stoichiometric principles to calculate the appropriate chemical equation for the reaction between iron metal and a solution of copper i sulfate introduction in the form of finely split powder copper metal is produced as a byproduct of the process as iron ions the ion meta dissolves

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web purpose to enhance the understanding of stoichiometry a reaction between iron and copper ii sulfate solution will be conducted this will help you to differentiate limiting and excess reactant in a chemical reaction finally the theoretical and percent yield of this reaction will be calculated

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web purpose in this experiment you will observe a single replacement reaction you will also use stoichiometry to predict what the theoretical yield of product is and calculate a percent yield reaction iron copper ii sulfate copper iron ii sulfate pre lab questions write the balanced equation

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web october 3 2017 abstract the techniques used in this lab are quantitative transfer and vacuum filtration with the reaction of 8 001 grams of copper ii sulfate CuSO_4 and 2 0153 grams of iron powder Fe the goal of this experiment was to determine the product of copper ii sulfate with iron

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web procedure 1 find the mass of a clean dry 100 ml beaker and record 2 measure out 8.00 grams of copper ii sulfate onto a piece of filter paper taking the filter paper's mass into account add these to the beaker 3 measure 50.0 ml of water in the cylinder and add it to the beaker 4

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web stoichiometry lab the reaction of iron with copper ii sulfate the study of stoichiometry deals with the calculation of quantities in a chemical reaction how much product will be produced how much reactant do you need to make that much product

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web description great lab for use during stoichiometry unit in chemistry reinforces ideas of mole ratios and stoichiometry calculations by measuring the mass of iron that reacts and the mass of copper metal produced students can calculate the ratio of moles of reactant to moles of product this experimental mole ratio from lab can be compared

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web in this experiment we will use stoichiometric principles to deduce the appropriate equation for the reaction between metallic iron and a solution of copper ii sulfate this reaction produces metallic copper which is seen precipitating as a finely divided red powder

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web in this lab we will use experimental evidence and the principles of stoichiometry to determine which ion is formed in a single replacement reaction between iron and copper ii sulfate pre lab assignment in preparation for the lab activity we will investigate another single replacement reaction between copper and silver nitrate

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web through this experiment we sought to use stoichiometric principles which derive from stoichiometry the study of substance quantities in ratio or proportion in a chemical reaction in this lab experiment the principles help distinguish the equation between iron powder and aqueous copper ii sulfate in the iron copper ii sulfate redox reaction

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web this experiment is focused on a reaction between iron metal and a solution of copper ii sulfate we are able to figure out the chemical equation between the two by using stoichiometric principles the iron displaces the copper ii sulfate ions the copper ii is then reduced and gains electrons

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web stoichiometry can be defined as the relationship between the relative quantities of substances taking part in a reaction

or forming a compound stoichiometric principles can be used to deduce the appropriate chemical equation for a reaction in our experiment we used solid iron in the form of powder was added to a solution of copper ii sulfate

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web stoichiometric principles can be used to determine the chemical equation between iron powder and aqueous copper ii sulfate in the iron copper ii sulfate redox reaction the iron ions dissolve in the copper ii sulfate solution to produce copper ii ions

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web under the finalization of the experiment it was found that iron was the limiting reactant and iron ii sulfate was a product of the reaction between 7 0002g away copper ii sulfate and 2 0101g iron metal which produced 2 3037g of copper product with a percent yield of 100 72 introduction

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web stoichiometry lab report the reaction of iron with copper ii sulfate period 5 in this experiment i observed the reaction of iron with copper ii sulfate i obtained the results to this experiment through adding heated CuSO_4 solution to the iron powder stirring the solution adding water then removing water adding acetone heating the acetone on low

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web stoichiometry lab iron and copper sulfate sample data data and observations mass of copper ii sulfate pentahydrate 3 27 g of g mass of filter paper mass of filter paper and dry copper mass of dry copper 1 28 g