

SALT ANALYSIS

TEST FOR ANIONS

EXPERIMENT	OBSERVATION	INFERENCE
1. Test for CO_3^{2-} → A little of the salt is treated with dil. HCl.	Brisk effervescence with the evolution of colourless odourless gas	May be CO_3^{2-} $\text{CaCO}_3 + 2\text{HCl} \rightarrow \text{CaCl}_2 + \text{CO}_2 + \text{H}_2\text{O}$
Confirmatory test :- 2. Pass the above gas through lime water.	Lime water turns milky	Presence of CO_3^{2-} is confirmed $\text{Ca(OH)}_2 + \text{CO}_2 \rightarrow \text{CaCO}_3 + \text{H}_2\text{O}$
Test for Halides 3. A little of the salt is treated with 1 or 2 drops of conc. H_2SO_4	a) Colourless pungent smelling gas evolves, dense white fumes when a glass rod is dipped in NH_4OH is shown on the mouth of the test tube b) Reddish brown vapours with pungent smell c) Deep violet vapours with pungent smell	a) May be Cl^- $\text{NaCl} + \text{H}_2\text{SO}_4 \rightarrow \text{NaHSO}_4 + \text{HCl}$ b) Maybe Br^- $\text{NaBr} + \text{H}_2\text{SO}_4 \rightarrow \text{NaHSO}_4 + \text{HBr}$ $\text{H}_2\text{SO}_4 + 2\text{HBr} \rightarrow \text{SO}_2 + \text{Br}_2 + 2\text{H}_2\text{O}$ c) May be I^- $\text{KI} + \text{H}_2\text{SO}_4 \rightarrow \text{KHSO}_4 + \text{HI}$ $\text{H}_2\text{SO}_4 + 2\text{HI} \rightarrow \text{SO}_2 + \text{I}_2 + 2\text{H}_2\text{O}$
Confirmatory test of halides 4. A little of the salt solution is acidified with dil. HNO_3 , boil, cool and add AgNO_3 sol.	a) Curdy white ppt. soluble in NH_4OH . b) Pale yellow ppt. sparingly soluble in NH_4OH c) Yellow ppt. insoluble in NH_4OH	a) presence of Cl^- is confirmed $\text{NaCl} + \text{AgNO}_3 \rightarrow \text{NaNO}_3 + \text{AgCl (white ppt.)}$ b) presence of Br^- is confirmed $\text{KBr} + \text{AgNO}_3 \rightarrow \text{KNO}_3 + \text{AgBr (pale yellow ppt.)}$ c) presence of I^- is confirmed $\text{KI} + \text{AgNO}_3 \rightarrow \text{KNO}_3 + \text{AgI (yellow ppt.)}$

Unknown Salt Identification

Phyllis Buell, James Girard



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College Practical Chemistry V K Ahluwalia, Sunita Dhingra, Sunita Dhingra, 2005 **ISC Practical Chemistry** Aditya Mitra, , **A Method for the Identification of Pure Organic Compounds by a Systematic Analytical Procedure Based on Physical Properties and Chemical Reactions ...** Samuel Parsons Mulliken, 1922 **Sif: Chemistry 5na Wb , Chemistry Lab Manual Class XI | follows the latest CBSE syllabus and other State Board following the CBSE Curriculum.** Mr. Rohit Manglik, Mr. Kaushalesh Dwivedi, 2022-08-04 With the NEP and expansion of research and knowledge has changed the face of education to a great extent In the Modern times education is not just constricted top the lecture method but also includes a practical knowledge of certain subjects This way of education helps a student to grasp the basic concepts and principles Thus trying to break the stereotype that subjects like Physics Chemistry and Biology means studying lengthy formulas complex structures and handling complicated instruments we are trying to make education easy fun and enjoyable **Breakthrough to CLIL for Chemistry Workbook** Richard Harwood, Timothy Chadwick, 2015-04-30 A series of workbooks offering integrated content and language support for specific subjects Breakthrough to CLIL for Chemistry Age 14 helps ESL EAL students get the most out of their studies when learning subjects through the medium of English The workbook contains exercises set within the context of core topics to consolidate understanding embedding practice in aspects of language central to the subject in question It is designed to support any Chemistry curriculum for students aged 14 16 including UK GCSE Cambridge IGCSE and IB MYP The book should be used alongside a core textbook and may be used within the classroom or as a self study or homework resource Salts and Their Reactions Leonard Dobbin, Hugh Marshall, 1904 Advanced Inorganic Analysis - I Mr. Ajeet Singh, 2018-05-22 This book provides in depth coverage of modern methods in inorganic analysis including advanced qualitative and quantitative techniques instrumentation and interpretation of results catering to postgraduate and research level chemistry students **Understanding 'O' Level Chemistry through Problem Solving ,** *Abiotic stress: molecular genetics and genomics, Volume II* Mukesh Jain, Rohini Garg, Rajeev K. Varshney, Prasanta Kumar Subudhi, **EduGorilla's CBSE Class 11th Chemistry Lab Manual | 2024 Edition | A Well Illustrated, Complete Lab Activity book with Separate FAQs for Viva Voce Examination ,** Need an informative and well illustrated Lab Manual CBSE Class 11th Chemistry Lab Manual is here for you The Lab Manual provides comprehensive steps for guiding students through each experiment Rigorously researched content prepared by a team of educators writers editors and proofreaders CBSE Class XI Chemistry Lab Manual has properly labeled high resolution diagrams and graphs A separate section on Viva Questions has been included to aid students in their Viva examination The Lab Manual explains the complex topics through detailed illustrations and lucid language making them simple to grasp Worksheets have been provided in CBSE Class 11th Chemistry Lab Manual for doing rough work **Modern Experimental Chemistry** George W. Jr. Latimer, 2012-12-02 Modern Experimental Chemistry provides techniques of qualitative analysis

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Laboratory Manual for Chemistry Fundamentals Phyllis Buell, James Girard, 2002-08 Physical Sciences *Laboratory Manual of Qualitative Analysis* Wilhelm Segerblom, 1908 **General, Organic, and Biochemistry Lab Manual** Sara Selfe, 2006-01-12 Offers a choice of classic chemistry experiments and innovative ones All of them place special emphasis on the biological implications of chemical concepts Available for custom publishing at <http://custompub.whfreeman.com>

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