

Unit 6: Molecules & Compounds

Key Terms

Ionic Bonding

Electron dot diagram
ion
anion
cation
chemical bond
ionic bond
chemical formula
crystals

Covalent Bonding

Covalent bond
molecule
polar covalent bond

Naming Compounds & Writing Formulas

Polyatomic ion

Essential Questions

Electron Configurations: **electron dot diagram** – a model of an atom in which each dot represents a valence electron.

PERIODIC TABLE ELEMENTS 1–20							
H ·							He ·
Li ·	Be ·	B ·	C ·	N ·	O ·	F ·	Ne ·
Na ·	Mg ·	Al ·	Si ·	P ·	S ·	Cl ·	Ar ·
K ·	Ca ·						

• What is a chemical bond?

A **chemical bond** is the force that holds atoms and ions together as a unit.

• How is an ionic bond different from a covalent bond?

Ionic Bonds: some elements achieve stable electron configurations through the transfer of electrons between atoms.



Formation of ions – when an atom gains or losses an electron, the number of protons is no longer equal to the number of electrons. The charge is no longer neutral.

ion – an atom with a net positive or negative charge. Represented by a + or – sign.

Cl gains an electron 17 protons 18 electrons charge of 1- Cl^{1-} or Cl^-

anion – (an eye un) an ion with a negative charge

Naming: using part of the element name plus the suffix *-ide*.

Cl^- : *Chloride* ion.

Na loses an electron 11 protons 10 electrons charge of 1+ Na^{1+} or Na^+

cation – (kat eye un) an ion with a positive charge

Naming: just use the elements name.

Unit 6 Study Guide Covalent Bonding

Thomas Elliott Taylor



Unit 6 Study Guide Covalent Bonding:

Study Guide and Solutions Manual for Seager/Slabaugh's Chemistry for Today Seager/Slabaugh, 2004 The fifth edition of the Study Guide and Student Solutions Manual has been updated to reflect all of the changes to the text This ancillary tests the student on the learning objectives in each chapter and provides answers to all of the even numbered end of chapter exercises New additional activities have been added to include a review of each section of the chapter and a section entitled Tying It All Together with a Laboratory Application

Life: The Science of Biology Study Guide William K. Purves, Edward Dzialowski, Lindsay Goodloe, Betty McGuire, Nancy Guild, Paula Mabee, 2003-12-26 New edition of a text presenting underlying concepts and showing their relevance to medical agricultural and environmental issues Seven chapters discuss the cell information and heredity evolutionary process the evolution of diversity the biology of flowering plants and of animals and ecology and biogeography Topics are linked by themes such as evolution the experimental foundations of knowledge the flow of energy in the living world the application and influence of molecular techniques and human health considerations Includes a CD ROM which covers some of the subject matter and introduces and illustrates 1 700 plus key terms and concepts Annotation copyrighted by Book News Inc Portland OR

Student Study Guide for Biology [by] Campbell/Reece Martha R. Taylor, 2002 Marty Taylor Cornell University Provides a concept map of each chapter chapter summaries a variety of interactive questions and chapter tests

Student Guide For Living Chemistry Barbara Ucko, 2012-12-02 Student Guide for Living Chemistry is a 23 chapter textbook guide that allows students to study and review on their own and test their understanding to help them prepare for examinations Every chapter begins with a list of objectives stating exactly the skills to develop in a particular unit Each objective corresponds to a section in the textbook Living Chemistry Three kinds of questions are provided for each objective to check the student s understanding namely short answer Study Questions multiple choice and fill in The answers for all questions are provided at the end of the chapter The opening chapters cover the SI units composition of matter chemical bonding compounds chemical change gases respiration and water The subsequent chapters deal with solutions acids bases salts nuclear and organic chemistry oxygen derivatives and hydrocarbons polymers and other organic derivatives This textbook also explores the chemistry of carbohydrates lipids proteins enzymes and energy and carbohydrate metabolism The remaining chapters discuss the chemistry of vitamins hormones body fluid drugs and poisons Undergraduate chemistry students will find this book invaluable

Study Guide to Accompany Basics for Chemistry Martha Mackin, 1982-01-01 Study Guide to Accompany Basics for Chemistry is an 18 chapter text designed to be used with Basics for Chemistry textbook Each chapter contains Overview Topical Outline Skills and Common Mistakes which are all keyed to the textbook for easy cross reference The Overview section summarizes the content of the chapter and includes a comprehensive listing of terms a summary of general concepts and a list of numerical exercises while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as

they appear in the textbook The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students

Study Guide for The Anatomy and Physiology Learning System Edith Applegate, 2010-03-30 Designed to accompany The Anatomy and Physiology Learning System 4th Edition by Edith Applegate this study guide helps you learn and review basic A P concepts Each chapter emphasizes medical terminology with a set of key terms word parts clinical terms and abbreviations and then adds a variety of fun filled learning exercises review questions a quiz and a word puzzle The study guide corresponds to the textbook chapter for chapter Chapter learning objectives help you focus on the most important material Key concepts are defined on the first page of each chapter in the workbook Learning exercises for each chapter include short answer matching and diagrams to label and color Self quizzes allow you to measure your progress and understanding Fun and Games features end each chapter with a variety of engaging puzzles covering words and concepts A chapter summary provides a brief review of each chapter A chapter review provides questions for reinforcement and review of the concepts in each chapter

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Study Guide for Chemistry by Steven S. Zumdahl Martha B. Barrett, 1986 **Inclusion Strategies That Work for Adolescent Learners!** Toby J. Karten, 2009-03-17 Wow What a wonderful resource for all teachers This book combines theory and practical strategies that can easily be implemented in anyone s classroom Kudos to the author Sarah N Miller Special Education Teacher Baldwin County Schools Summerdale AL This book will quickly become the must have resource for all special and general educators Karten addresses all aspects of the inclusive environment beginning with the inclusive mind set and working through environment structure content and most important the idiosyncratic adolescent Harold M Tarriff Director of Special Services School District of the Chathams NJ Strategies to achieve winning results in the inclusive secondary classroom Higher performance and more positive experiences are possible for all adolescent learners with some guidance perseverance and the right techniques Toby J Karten provides teachers with a practical approach for creating a

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Study Guide to Accompany Introduction to Chemistry J.C. Cummins, Jack D. Cummins, 1975 Study Guide for Mortimer's Chemistry, Sixth Edition Donald W. Shive, Louise E. Shive, 1986 *Science Indiana Standards Manager Grade 6* McDougal Littell, 2004

General Chemistry Study Guide Sixth Edition Darrell D. Ebbing, 1999 **Study Guide for Chemical Principles** Thomas Elliott Taylor, 1979 *Cell and Molecular Biology, Problems Book and Study Guide* Gerald Karp, 2001-09-25 Balances coverage of the concepts of cell and molecular biology using examples of experimentation to support those concepts As experimental techniques become more diverse and complex it is increasingly necessary to identify individual studies that have a broad impact on our understanding of cell biology This text describes in detail some of the key experimental findings along with the original data and figures *Study Guide for Biochemistry, 2nd Ed., [by] Christopher K. Mathews, K.E. Van Holde* Christopher K. Mathews, Harry Ako, 1996 **The Sciences, Study Guide** James Trefil, Anthony J. Gaudin, Robert M. Hazen, 2000-11-24 The Sciences An Integrated Approach 3e is the only text on the market for the integrated science course It integrates physics chemistry astronomy earth sciences and biology for students with little or no science background The new third edition builds on the successful previous edition by incorporating changes that meet reform curriculum guidelines Given the dynamic nature of the science disciplines covered in Trefil's text updated content is critical The new third edition has thoroughly updated content throughout bringing the most up to date discoveries to the students in all five disciplines its broad approach that covers such a variety of the science disciplines updated content is crucial This revision has several significant changes URL's are included at the ends of each chapter many of the Science News boxes are updated to show hot topics *Inquiry Into Life, Study Guide* Sylvia S. Mader, 1982 **Study Guide to accompany The Sciences: An Integrated Approach, 4th Edition** James Trefil, Robert M. Hazen, 2003-05-22 Work more effectively and gauge your progress along the way Designed to be used alongside Trefil's The Sciences 4th Edition this Study Guide contains many elements that foster student success Included are chapter reviews learning objectives key chapter concepts and key concept charts The ties between science and math are reinforced with key formulas and equations Links to scientists and their

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