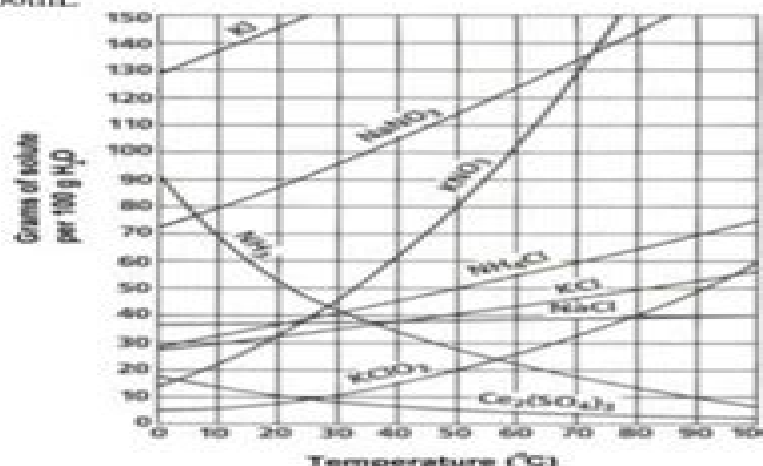


Name: _____ Date: _____ Period: _____

Solubility Curve Practice Problems

Directions: Find the mass of solute will dissolve in 100mL of water at the following temperatures?

1. KNO_3 at 70°C = _____
2. NaCl at 100°C = _____
3. NH_4Cl at 90°C = _____
4. Which of the **above** three substances is most soluble in water at 15°C . = _____
5. Which of the **above** three substances is least soluble in water at 15°C . = _____



2. Solution	Saturated, super saturated or Unsaturated?	If unsaturated: How much more solute can dissolve in the solution?
a solution that contains 70g of NaNO_3 at 30°C (in 100 mL H_2O)		
a solution that contains 50g of NH_4Cl at 50°C (in 100 mL H_2O)		
a solution that contains 20g of KClO_3 at 50°C (in 50 mL H_2O)		
a solution that contains 70g of KI at 0°C (in 200 mL H_2O)		

3. a. How many grams of KCl can dissolve at 5°C ? _____
 b. How many grams of KCl can dissolve at 25°C ? _____
 c. How many grams of $\text{Ce}_2(\text{SO}_4)_3$ can dissolve at 10°C ? _____
 d. How many grams of $\text{Ce}_2(\text{SO}_4)_3$ can dissolve at 50°C ? _____
4. At 90°C , you dissolved 10 g of KCl in 100. g of water. Is this solution saturated or unsaturated?
5. A mass of 100 g of NaNO_3 is dissolved in 100 g of water at 80°C .
 a) Is the solution saturated or unsaturated? _____
 b) As the solution is cooled, at what temperature should solid first appear in the solution? Explain.
6. Use the graph to answer the following two questions:
 Which compound is most soluble at 20°C ?
 Which is the least soluble at 40°C ?
 Which substance on the graph is **least** soluble at 10°C ?

Unit 12 Solutions Solubility Curves Answer Key

M Carnoy



Unit 12 Solutions Solubility Curves Answer Key:

Unit 12 Solutions Solubility Curves Answer Key Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become much more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Unit 12 Solutions Solubility Curves Answer Key**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve to the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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