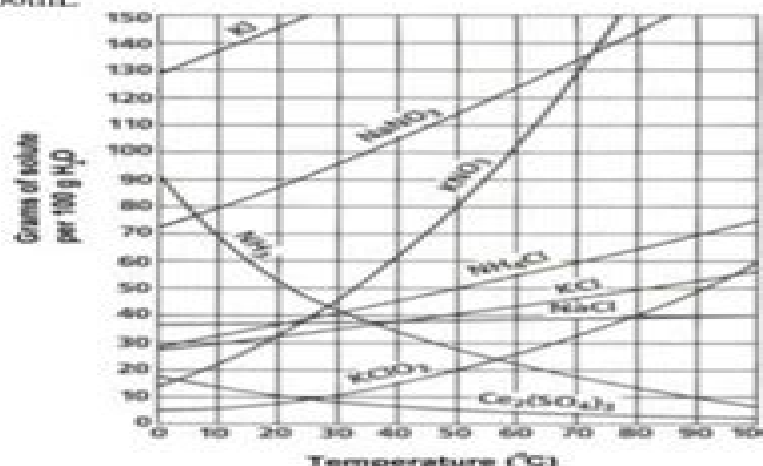


Name: _____ Date: _____ Period: _____

Solubility Curve Practice Problems

Directions: Find the mass of solute will dissolve in 100mL of water at the following temperatures?

1. KNO_3 at 70°C = _____
2. NaCl at 100°C = _____
3. NH_4Cl at 90°C = _____
4. Which of the **above** three substances is most soluble in water at 15°C . = _____
5. Which of the **above** three substances is least soluble in water at 15°C . = _____



2. Solution	Saturated, super saturated or Unsaturated?	If unsaturated: How much more solute can dissolve in the solution?
a solution that contains 70g of NaNO_3 at 30°C (in 100 mL H_2O)		
a solution that contains 50g of NH_4Cl at 50°C (in 100 mL H_2O)		
a solution that contains 20g of KClO_3 at 50°C (in 50 mL H_2O)		
a solution that contains 70g of KI at 0°C (in 200 mL H_2O)		

3. a. How many grams of KCl can dissolve at 5°C ? _____
 b. How many grams of KCl can dissolve at 25°C ? _____
 c. How many grams of $\text{Ce}_2(\text{SO}_4)_3$ can dissolve at 10°C ? _____
 d. How many grams of $\text{Ce}_2(\text{SO}_4)_3$ can dissolve at 50°C ? _____
4. At 90°C , you dissolved 10 g of KCl in 100. g of water. Is this solution saturated or unsaturated?
5. A mass of 100 g of NaNO_3 is dissolved in 100 g of water at 80°C .
 a) Is the solution saturated or unsaturated? _____
 b) As the solution is cooled, at what temperature should solid first appear in the solution? Explain.
6. Use the graph to answer the following two questions:
 Which compound is most soluble at 20°C ?
 Which is the least soluble at 40°C ?
 Which substance on the graph is **least** soluble at 10°C ?

Unit 12 Solutions Solubility Curves Answer Key

**Neale Taylor, Angela Stubbs, Robert
Stokes**



Unit 12 Solutions Solubility Curves Answer Key:

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and Multidisciplinary and Applied Science They are also grouped by type core materials supplementary units and science activity books Each annotation of curriculum material includes a recommended grade level a description of the activities involved and of what students can be expected to learn a list of accompanying materials a reading level and ordering information The curriculum materials included in this book were selected by panels of teachers and scientists using evaluation criteria developed for the guide The criteria reflect and incorporate goals and principles of the National Science Education Standards The annotations designate the specific content standards on which these curriculum pieces focus In addition to the curriculum chapters the guide contains six chapters of diverse resources that are directly relevant to middle school science Among these is a chapter on educational software and multimedia programs chapters on books about science and teaching directories and guides to science trade books and periodicals for teachers and students Another section features institutional resources One chapter lists about 600 science centers museums and zoos where teachers can take middle school students for interactive science experiences Another chapter describes nearly 140 professional associations and U S government agencies that offer resources and assistance Authoritative extensive and thoroughly indexed and the only guide of its kind Resources for Teaching Middle School Science will be the most used book on the shelf for science teachers school administrators teacher trainers science curriculum specialists advocates of hands on science teaching and concerned parents Holt Chemistry Ralph Thomas Myers,2004

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