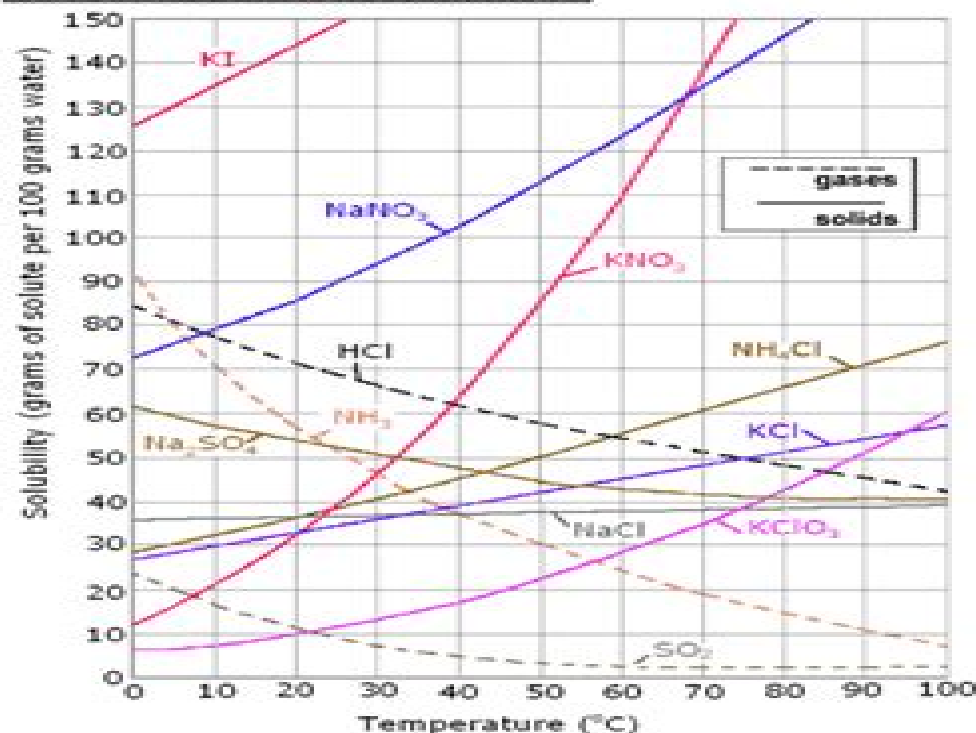


UNIT 12 - SOLUTIONS

SOLUBILITY CURVES WORKSHEET



- Which compound is *least* soluble at 20 °C? At 80 °C?
- Which substance is the *most* soluble at 10 °C? At 50 °C? At 90 °C?
- The solubility of which substance is *most* affected by changes in temperature?
- The solubility of which substance is *least* affected by changes in temperature?
- Are the following solutions saturated, unsaturated, or supersaturated?
 (Assume all are dissolved in 100 grams of water.)
 (A) 50 grams of KNO₃ at 50 °C
 (B) 100 grams of NaNO₃ at 80 °C
 (C) 30 grams of KNO₃ at 25 °C
 (D) 50 grams of KCl at 80 °C
 (E) 65 grams of NH₄Cl at 70 °C
 (F) 90 grams of KNO₃ at 60 °C
- NH₃ is a gas. Describe what happens to its solubility as the temperature goes from 20 °C to 80 °C.

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American Chemical Society

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Table of Contents Unit 12 Solutions Solubility Curves

1. Understanding the eBook Unit 12 Solutions Solubility Curves
 - The Rise of Digital Reading Unit 12 Solutions Solubility Curves
 - Advantages of eBooks Over Traditional Books
2. Identifying Unit 12 Solutions Solubility Curves
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Unit 12 Solutions Solubility Curves
 - User-Friendly Interface
4. Exploring eBook Recommendations from Unit 12 Solutions Solubility Curves
 - Personalized Recommendations
 - Unit 12 Solutions Solubility Curves User Reviews and Ratings
 - Unit 12 Solutions Solubility Curves and Bestseller Lists
5. Accessing Unit 12 Solutions Solubility Curves Free and Paid eBooks
 - Unit 12 Solutions Solubility Curves Public Domain eBooks
 - Unit 12 Solutions Solubility Curves eBook Subscription Services
 - Unit 12 Solutions Solubility Curves Budget-Friendly Options
6. Navigating Unit 12 Solutions Solubility Curves eBook Formats

- ePub, PDF, MOBI, and More
- Unit 12 Solutions Solubility Curves Compatibility with Devices
- Unit 12 Solutions Solubility Curves Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Unit 12 Solutions Solubility Curves
 - Highlighting and Note-Taking Unit 12 Solutions Solubility Curves
 - Interactive Elements Unit 12 Solutions Solubility Curves
- 8. Staying Engaged with Unit 12 Solutions Solubility Curves
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Unit 12 Solutions Solubility Curves
- 9. Balancing eBooks and Physical Books Unit 12 Solutions Solubility Curves
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Unit 12 Solutions Solubility Curves
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Unit 12 Solutions Solubility Curves
 - Setting Reading Goals Unit 12 Solutions Solubility Curves
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Unit 12 Solutions Solubility Curves
 - Fact-Checking eBook Content of Unit 12 Solutions Solubility Curves
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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