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Unit 6 – Worksheet 4 Molecular Compounds

1

Properties.

Basic structural unit

- Name each of the following binary compounds of non-metallic elements
- a CBr4
- b. N₂P₃
- c. PCl₃
- d: ICI
- e. N2^O
- f SiF4
- g GeH₄
- h N₂Br₄
- i. P₂S₅
- j SeO2
- k NH3
- 1.802

- Write the formula for the following binary compounds of nonmetallic elements
- a. phosphorus triiodide
- b. silicon tetrachloride
- c. dinitrogen pentoxide
- e dinitrogen tetroxide
- f. carbon monoxide
- g carbon dioxide
- h sulfur hexafluoride
- i dinitrogen tetrachloride
- j. carbon tetrasodide
- k phosphorus pentafluoride
- 1 diphosphorus pentoxide

Unit 6 4 Molecular Compounds Answers

Greg Curran

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