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 $2\text{NaCl} + \text{F}_2 \rightarrow 2\text{NaF} + \text{Cl}_2$
 $2\text{H}_2\text{O} + 5\text{C} \rightarrow 2\text{H}_2\text{O} + 5\text{C}$
 $\text{Pb} + \text{OH}^- \rightarrow \text{PbCl}_2 + 6\text{H}^+$
 $3\text{Al} + 3\text{K}_2\text{SO}_4 \rightarrow 4\text{KBr} + \text{Al}_2\text{SO}_4 + 7\text{H}_2\text{O}$
 $4\text{H}_2\text{O} + 2\text{CO}_2 \rightarrow 2\text{H}_2\text{O} + 8\text{H}^+$
 $3\text{H}_2\text{O} + 5\text{O}_2 \rightarrow 3\text{CO}_2 + 4\text{H}_2\text{O}$

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