

Activity 1.3.3 Thermodynamics Answer Key

Introduction

Think back to the last time someone complained about a door being left open. What did you notice about the temperature within the room as a result of the open door? In Activity 1.3.3 you will investigate the effects of work, thermal energy, and energy on a system, as in the case of the room with the door left open.

Procedure

Answer the following questions as your teacher discusses the Introduction to Thermodynamics presentation.

1. Define thermodynamics.

The study of the effects of work, thermal energy, and energy on a system

2. List three examples of a thermodynamic system.

- a. **Nuclear power**
- b. **Electronic heat sink**
- c. **Rocket launch**

3. Define thermal energy.

Kinetic energy in transit from one object to another due to temperature difference

4. Define temperature.

The average kinetic energy of particles in an object

5. Fill in the table below with the correct scale and unit.

Scale	Freezing point of water	Boiling point of water
Celsius	0°C	100°C
Fahrenheit	32°F	212°F
Kelvin	273K	373K

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