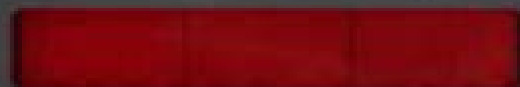




Westinghouse IQ Data Plus II

Alarm Trip



Units
Kilo-Units
Mega-Units

- I_A Amps RMS
- I_B Amps RMS
- I_C Amps RMS
- V_{A-B} Volts RMS
- V_{B-C} Volts RMS
- V_{C-A} Volts RMS
- V_{A-N} Volts RMS
- V_{B-N} Volts RMS
- V_{C-N} Volts RMS
- Watts
- Vars
- Power Factor
- Demand Watts
- Frequency
- Watt-Hours

Reset

RES

Step
Display

Up



Down



ALARM CODES

1. EXTERNAL TRIP
2. OVERVOLTAGE
3. UNDERVOLTAGE
4. PHASE UNBALANCE
5. PHASE LOSS
6. PHASE REVERSAL
7. MALFUNCTION

Westinghouse Iq Data Plus Ii Manual

Benjamin Stein, John S. Reynolds



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web $m \times c \times \Delta t$ equation 1 where m is the total mass of the solution solute plus solvent c is the specific heat of the solution and Δt is the observed temperature change the specific heat of the solution is generally assumed to be the same as that of *guided inquiry designing the ideal hand warmer* - Dec 27 2022

web the heat of reaction ΔH_{soln} is written after the products in units of kJ mol rxn for example sodium hydroxide dissolves exothermically releasing 44.2 kilojoules per mole dissolved this process is represented as $\text{NaOH (s)} \rightarrow \text{NaOH (aq)}$ $\Delta H_{\text{soln}} = -44.2 \text{ kJ mol rxn}$ write equations to similarly represent the dissolving process for

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web the goal of my experiment and research is to evaluate the various number of compounds to be used as hand warmers and locate the ones that can raise the temperature to 20 degrees celsius in 100 g water and costs as little as possible

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web in this lab you are challenged to use chemistry to determine which ionic compound makes the best hand warmer in this experiment you will collect data that will allow you to calculate the heat of solution ΔH_{soln} or the amount of energy released or absorbed in kJ/mol of the substance being dissolved

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web mar 4 2014 contains water the hand warmer is struck in a manner that ruptures the inner pouch releasing the ionic salt into the water of the outer pouch the salt dissolves and the water warms this type of hand warmer tends to produce a more vigorous heat than the dry powder type of hand warmer but does not produce heat for quite as long

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