

Grade 7 Wellness Unit 2 Performance Task Rubric: student led lesson/magazine article

Criteria	Exemplary (E) 94%	Preficient (PR) 88%	Developing (DE) 76%
	A-	B+	C+
Critical thinking/ Problem solving <i>10 points</i>	☐ Asks thought-provoking essential questions that elaborate on the problem. Asks extra questions that involve the audience. ☐ Brainstorms and evaluates different problem-solving strategies and prioritizes them in order to choose the most effective strategy. ☐ The audience is left with a clear sense of empowerment ("aha moment") and deeper understanding. ☐ Proposes a variety of creative solutions/hypotheses that indicate a deep comprehension of the problem.	☐ Asks probing essential questions that explore the problem. ☐ Brainstorms a few strategies without prompting. ☐ The work is informative; it contains useful information presented in an accessible form. ☐ Proposes a variety of solutions/hypotheses that indicate general comprehension of the problem.	☐ Asks general questions about parts of the problem. ☐ Brainstorms a few simple strategies with prompting. ☐ The work is somewhat informative; it contains material that is somewhat superficial. ☐ Proposes a limited number of solutions/hypotheses that may not address all the issues of the problem.
Comments:			
Communication/ Language <i>10 points</i>	☐ Student lesson is thought provoking, well-paced, engaging and highly-effective. Audience is involved. ☐ Article communicates the message in an exceptionally creative way including a clear explanation and thought-provoking message. ☐ Uses fluent, original and highly appropriate language that is precise and engaging, with notable sense of voice, awareness of audience and purpose and varied sentence structure. ☐ MLA formatted bibliography included in the project (5 or more sources) (at least one is printed material).	☐ Student lesson is effective and thoughtful. ☐ Article communicates the message in a creative way with a message that needed further development. ☐ Uses language that is fluent and original, with a sense of voice, awareness of audience and purpose, and the ability to vary sentence structure. ☐ Bibliography present (3-5 sources); Some errors in the formatting or not included as part of the project.	☐ Student lesson is lacking thought and preparation. ☐ Article needs more thought and appears rushed. ☐ Use basic but appropriate language, with a basic sense of voice, some awareness of audience and purpose and some attempt to vary sentence structure. ☐ Bibliography not MLA format or absent from the project.
Comments:			
Connection and Collaboration (empathy) <i>10 points</i>	☐ Mature group: disciplined; able to see and feel what others see and feel; unusually open to and willing to seek out the odd, alien, or different points of view and express them. ☐ All the tasks were shared out effectively and cooperatively to the group members equally. Stayed actively focused for the duration of the task. Listened to each other viewpoints. ☐ Task is completed on time and they were well-prepared and it was professional.	☐ Sensitive: see and feel what others see and feel; open to the unfamiliar or different points of view but found it difficult to express them. ☐ All the tasks were shared out to the group members but not always equally. Stayed on task throughout. ☐ Task is completed on time but needed time to prepare the group.	☐ Awake: knows and feels that others see and feel differently, and is somewhat able to empathize with others. Need support and guidance from a teacher. ☐ Group needed help and support to divide the tasks and stay on task. ☐ Task was not completed on time.
Comments:			
Self Reflection:An important lifelong lesson that I can take away is _____. When working on a team, I am good at _____. My strengths as a learner are _____. Some areas for improvement include _____.			

Unit 2 Wild West Performance Task

Huangqi Zhang



Unit 2 Wild West Performance Task:

America's Story, 1997 **Monthly Catalogue, United States Public Documents**, 1994 **Government Reports Announcements & Index**, 1988 *Clover Units 1-2 Coal-fired Generating Station and Related Transmission Facilities Construction and Operation, Old Dominion Electric Cooperative, Halifax County*, 1990 *Monthly Catalog of United States Government Publications*, 1994 **Resources in Education**, 1998 **Interim Assessment: Emissions and control**, 1987 **Special Education in Today's Diverse Classrooms** Shantel M. Farnan, Ruby L. Owiny, 2025-09-02

Special Education in Today's Diverse Classrooms Meeting the Needs of Students with Exceptionalities is an introductory level textbook designed for all pre service teachers to learn about meeting the needs of students with exceptionalities in inclusive environments Along with descriptions of each IDEA disability category the book presents high leverage practices HLPs and evidence based strategies that are practical and applicable to any instructional environment Through the lens of HLPs this text emphasizes universal design for learning UDL tiered supports culturally inclusive pedagogies and practices CIPP and evidence based practices EBPs This textbook bridges the gap between research knowledge about disabilities and a practical approach to educating students offering a comprehensive framework for educators navigating the diverse needs of students with exceptionalities By placing a strong emphasis on CIPP and EBPs as they relate to HLPs it equips readers with tools to create meaningful and equitable learning experiences The unique structure enriched by authentic vignettes and aligned with professional standards ensures the practical application of frameworks such as UDL and multi tiered systems of support Additionally the book underscores the importance of family engagement making it a vital resource for fostering collaboration in education The content aligns with the Council for Exceptional Children CEC standards ensuring its relevance and utility for professional educator preparation Through its innovative approach this text inspires educators to not only meet students needs but also celebrate their individuality preparing them to succeed in dynamic inclusive school and classroom environments Key Features Real life vignettes from individuals with disabilities their families and educators offer authentic perspectives that go beyond case studies Links to resources to increase exposure and knowledge about specific topics designed to enrich understanding and application of inclusive practices Includes coverage of concepts such as trauma neurodiversity social and emotional learning assistive technology and new instructional technologies With inclusive language and culturally inclusive pedagogies and practices the book prepares future educators to foster trust and promote equity in their classrooms Focus on application to the classroom through questions and activities at the end of each chapter Color graphics visual frameworks e g UDL models and instructional charts enhance comprehension and engagement Please note ancillary materials such as quizzes and eFlashcards are not available as in the print version of this work **Interim Assessment**, 1987 **Subject Index to Unclassified ASTIA Documents** Defense Documentation Center (U.S.), 1960

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web answer $\text{Na}^+ \text{aq}$ and $\text{SO}_4^{2-} \text{aq}$ identify the chemical equation the spectator ions $\text{KOH} \text{aq}$ $\text{HNO}_3 \text{aq}$ $\text{KNO}_3 \text{aq}$ H_2O $\text{K}^+ \text{aq}$ $\text{OH}^- \text{aq}$ $\text{H}^+ \text{aq}$ $\text{NO}_3^- \text{aq}$ $\text{K}^+ \text{aq}$ $\text{NO}_3^- \text{aq}$ H_2O answer $\text{K}^+ \text{aq}$ and $\text{NO}_3^- \text{aq}$ identify the chemical equation the spectator ions $\text{Pb(NO}_3)_2 \text{aq}$ $2\text{KI} \text{aq}$ $\text{PbI}_2 \text{s}$ $2\text{KNO}_3 \text{aq}$

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web these ions are called spectator ions because they do not participate in the actual reaction canceling the spectator ions gives the net ionic equation which shows only those species that participate in the chemical reaction $2\text{Ag}^+ \text{aq} + \text{CrO}_4^{2-} \text{aq} \rightarrow \text{Ag}_2\text{CrO}_4 \text{s}$

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web chemical equation in which all dissolved ionic reactants and products including spectator ions are explicitly represented by formulas for their dissociated ions molecular equation chemical equation in which all reactants and products are

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web jan 29 2020 the ionic form of this reaction is $2\text{Na}^+ \text{aq} + 2\text{Cl}^- \text{aq} + \text{Cu}^{2+} \text{aq} + \text{SO}_4^{2-} \text{aq} \rightarrow 2\text{Na}^+ \text{aq} + \text{SO}_4^{2-} \text{aq} + \text{CuCl}_2 \text{s}$ the sodium ions and sulfate ion are the spectator ions in this reaction they appear unchanged in both the product and reactant side of the equation

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web mar 11 2014 solution ionic equation $\text{H}^+ \text{aq} + \text{Cl}^- \text{aq} + \text{Na}^+ \text{aq} + \text{HCO}_3^- \text{aq} \rightarrow \text{Na}^+ \text{aq} + \text{Cl}^- \text{aq} + \text{H}_2\text{O} \text{l} + \text{CO}_2 \text{g}$ the spectator ions are Cl^- and Na^+ net ionic equation $\text{H}^+ \text{aq} + \text{HCO}_3^- \text{aq} \rightarrow \text{H}_2\text{O} \text{l} + \text{CO}_2 \text{g}$ here is a video on spectator ions

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web the six $\text{NO}_3^- \text{aq}$ ions and the six $\text{Na}^+ \text{aq}$ ions that appear on both sides of the equation are spectator ions that can be canceled to give the net ionic equation $3\text{Ba}^{2+} \text{aq} + 2\text{PO}_4^{3-} \text{aq} \rightarrow \text{Ba}_3(\text{PO}_4)_2 \text{s}$

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web these spectator ions whose presence is required to maintain charge neutrality are neither chemically nor physically changed by the process and so they may be eliminated from the equation to yield a more succinct representation called a

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 $\text{AgNO}_3 \text{ aq} + \text{KCl aq} \rightarrow \text{AgCl s} + \text{KNO}_3 \text{ aq}$ total ionic $\text{Ag}^+ \text{ aq} + \text{NO}_3^- \text{ aq} + \text{K}^+ \text{ aq} + \text{Cl}^- \text{ aq} \rightarrow \text{AgCl s} + \text{K}^+ \text{ aq} + \text{NO}_3^- \text{ aq}$ net ionic $\text{Ag}^+ \text{ aq} + \text{Cl}^- \text{ aq} \rightarrow \text{AgCl s}$

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web video answer you know spectator ions are the ions present on the both sides of the equation indicate cancels out in the total ionic equations so you know in a case chemical reactions are described with chemical equations which symbolically present the starting materials end products

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web the equation below is the final net ionic equation after elimination of the spectator ions which yields water $\text{H}^+ \text{ aq} + \text{OH}^- \text{ aq} \rightarrow \text{H}_2\text{O l}$ the spectator ions in KOH aq and $\text{HNO}_3 \text{ aq}$ are $\text{K}^+ \text{ aq}$ and $\text{NO}_3^- \text{ aq}$ how to identify and list the

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web apr 6 2022 spectator ion a spectator ion is an ion that exists as a reactant and a product in a chemical equation a spectator ion can therefore be observed in the reaction of aqueous solutions of sodium carbonate and copper ii sulfate but does not affect the equilibrium what are spectator ions

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web the $\text{Ag}^+ \text{ aq}$ and $\text{Cl}^- \text{ aq}$ ions become AgCl s but the $\text{Na}^+ \text{ aq}$ ions and the $\text{NO}_3^- \text{ aq}$ ions stay as $\text{Na}^+ \text{ aq}$ ions and $\text{NO}_3^- \text{ aq}$ ions these two ions are examples of spectator ions ions that do nothing in the overall course of a chemical reaction

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web aug 8 2022 a spectator ion is an ion that does not take part in the chemical reaction and is found in solution both

before and after the reaction the net ionic equation is the chemical equation that shows only those elements compounds and ions that are directly involved in the chemical reaction

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web the new format requires students to write balanced chemical equations showing only the reacting substances excluding spectator ions for three different sets of reactants and to answer a short question requiring no calculator about each reaction the questions are intended to test the students understanding of the meaning of the reactions

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web jun 2 2023 the spectator ions are cut out from a reaction hence the net ionic equation for the above reaction would be $2\text{Cl}^- \text{aq} + \text{Cu}^{2+} \text{aq} \rightarrow \text{Cu}^{2+} \text{aq} + 2\text{Cl}^- \text{aq}$ these spectator ions are not actively involved in the reaction the existence of spectator ions affects the debye length

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