

Chem. 145, Problem Set, Chapter 1.

Marathon Problem:

Adapted from Zumdahl Chemistry, 1993. Page 39

A cylindrical bar of gold that is 1.5 in high and 0.25 in in diameter has a mass of 23.1964 g, as determined on an analytical balance. An empty graduated cylinder is weighed on a triple beam balance and has a mass of 73.47 g. After pouring a small amount of liquid into the graduated cylinder, the mass is 79.16 g. When the gold cylinder is placed in the graduated cylinder (the liquid covers the top of the gold cylinder), the volume indicated on the graduated cylinder is 8.5 mL. Assume that the temperatures of the gold bar and the liquid are 86°F.

What is the density of the gold at 30°C?

What is the density of the liquid at 86°F?

What is the density of the liquid at 300 K?

The first part of answering this question is to determine what information is required, and decide how you want to solve the question. Start with finding the density of the gold at 30 °C. What information do you need?

Recall:

$$\text{density} = \frac{\text{mass}}{\text{volume}}$$

Since you are looking for the density, you need the mass and the volume to solve the problem.

The mass is easy, that is given in the problem:

$$\text{mass}_{\text{gold}} = 23.1964 \text{ gm} \quad (\text{NOTE: gm stands for gram in Mathematica.})$$

So next we need the volume. But, first what is the point of the temperature? Is mass constant with temperature? Is volume constant with temperature? Think about some examples in everyday life. Why does a Hot Air Balloon work? Why do they place expansion joints in the highway?

Yes, volume changes with temperature. So, first what temperature is the gold bar in the problem?

A cylindrical bar of gold that is 1.5 in high and 0.25 in in diameter.... Assume that the temperatures of the gold bar and the liquid are 86°F.

So, is 86 °F the same as 30 °C?

$$T_{\text{temp}_F} = 86$$

$$T_{\text{temp}_C} = (T_{\text{temp}_F} - 32) \cdot \frac{5}{9}$$

$$T_{\text{temp}_C} = 30$$

Zumdahl Marathon Problems Answers

Steven S Zumdahl, Susan A Zumdahl



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Table of Contents Zumdahl Marathon Problems Answers

1. Understanding the eBook Zumdahl Marathon Problems Answers
 - The Rise of Digital Reading Zumdahl Marathon Problems Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Zumdahl Marathon Problems Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Zumdahl Marathon Problems Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Zumdahl Marathon Problems Answers

- Personalized Recommendations
- Zumdahl Marathon Problems Answers User Reviews and Ratings
- Zumdahl Marathon Problems Answers and Bestseller Lists
- 5. Accessing Zumdahl Marathon Problems Answers Free and Paid eBooks
 - Zumdahl Marathon Problems Answers Public Domain eBooks
 - Zumdahl Marathon Problems Answers eBook Subscription Services
 - Zumdahl Marathon Problems Answers Budget-Friendly Options
- 6. Navigating Zumdahl Marathon Problems Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Zumdahl Marathon Problems Answers Compatibility with Devices
 - Zumdahl Marathon Problems Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Zumdahl Marathon Problems Answers
 - Highlighting and Note-Taking Zumdahl Marathon Problems Answers
 - Interactive Elements Zumdahl Marathon Problems Answers
- 8. Staying Engaged with Zumdahl Marathon Problems Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Zumdahl Marathon Problems Answers
- 9. Balancing eBooks and Physical Books Zumdahl Marathon Problems Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Zumdahl Marathon Problems Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Zumdahl Marathon Problems Answers
 - Setting Reading Goals Zumdahl Marathon Problems Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Zumdahl Marathon Problems Answers

- Fact-Checking eBook Content of Zumdahl Marathon Problems Answers
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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