

## Unit 2 Exam Study Guide

*Use this guide to direct your attention toward the information to be covered by the test.*

### Biochemistry (Chapter 2)

1. Which type of energy moves matter?  
**Kinetic Energy**
2. Which elements make up around 96% of the body's mass?  
**Carbon, Oxygen, Hydrogen, Nitrogen**
3. What is the most abundant element in the human body?  
**Carbon**
4. Which element is the simplest atom?  
**Hydrogen**
5. Where are the three subatomic particles that make up the atom located?  
**Proton and Neutrons: Nucleus of Atom**  
**Electrons: outside of Nucleus**
6. What are the electrical charges of the subatomic particles?  
**Protons: +**  
**Neutrons: No charge**  
**Electrons: -**
7. What happens when sodium reacts with chlorine to form salt?  
**Ionic Bond**
8. Which type of bond is formed when electrons are shared equally?  
**Covalent Bond**
9. How does a compound differ from a molecule?  
**Molecule: two or more of the same atoms binds together**  
**Compound: two or more different atoms bind together**
10. What is an example of a synthesis reaction?  
 **$A + B \rightarrow AB$**
11. Why is water useful in body processes?  
**High Heat Capacity**  
**Polarity/Solvent Properties**  
**Chemical reactivity**  
**Cushioning**
12. Which numbers on the pH scale indicate an acid?  
**0-6.9**
13. What is the definition of an inorganic molecule?  
**Molecules that do not have repeating carbon chains**
14. Which type of compound is starch?  
**Carbohydrate (organic)**
15. Which organic compounds serve as enzymes in catalyzing reactions?  
**Proteins**
16. How do the nitrogenous bases of DNA pair up when in the double helix?  
**Adenine-Thymine**  
**Guanine-Cytosine**
17. How would you describe the function of enzymes?  
**Acts as a catalyst by lowering activation energy**
18. Which element is present in proteins and nucleic acids but not in carbohydrates or lipids?  
**Nitrogen**
19. How does the structure of RNA differ from that of DNA?  
**DNA is double stranded, RNA is single Stranded**  
**RNA contains Uracil, DNA contains thymine**
20. Be able to identify the four major classes of organic compounds by shape.  
**Check notes for structures**
21. What is the outermost shell of electrons?  
**Valence Electrons**
22. Which word describes molecules that make up living matter but do not contain carbon?  
**Inorganic**
23. What are some examples of monosaccharides?  
**Glucose, galactose**
24. What is the energy compound that supplies usable energy to all our cells?  
**ATP**
25. Be familiar with the properties of water that are important to living things.  
**High Heat capacity: Water can absorb a lot of heat**  
**Universal Solvent**  
**Polar: uneven distribution of charges**

# Unit 2 Study Guide Answers

**Carmela M. Krueser**



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