

Types of Reactions Lab



Name _____

Date _____ Period _____

INTRODUCTION

Matter can undergo both physical and chemical changes. Chemical changes result in the formation of new materials. Observable signs of chemical reaction include the release of a gas, a change in color, the formation of a solid, and the production of heat and light.

Chemical reactions can be classified in many ways. One classification system involves five general types of reactions: synthesis, decomposition, single displacement, double displacement, and combustion. In this lab, you will perform several reactions and classify them as one or more of the five general types.

SAFETY

- When heating a test tube, aim the mouth of a test tube AWAY from yourself and others.
- Do NOT look directly at burning magnesium. The intense light may damage your eyes.

PROCEDURE

1. Create a table of observations that includes 5 columns: Reactant(s), Appearance of Reactant(s) Before Mixing, Mass of Reactant(s), Appearance of Product(s), Mass of Product(s).

Reaction A

2. Obtain 5 mL of sodium sulfate (Na_2SO_4) solution in a clean test tube. Next, obtain 5 mL of barium chloride (BaCl_2) solution in another test tube. Place both test tubes in a beaker and measure the mass of the beaker with its contents. Record your observations. Pour the contents from one test tube into the other. Place both test tubes back into the beaker and measure the mass of the beaker once again. Record your observations.

Reaction B

3. Obtain 5 mL of copper(II) sulfate solution in a clean test tube. Obtain a small piece of steel wool (iron). Place the steel wool into a beaker, and the test tube into the same beaker. Measure the mass of the beaker with its contents. Record your observations. Place the steel wool into the test tube and observe the reaction. Place the test tube back into the beaker and measure the mass of the beaker once again. Record your observations.

Reaction C

4. Obtain 5 mL of hydrogen peroxide (H_2O_2) in a clean test tube. Use a scoopula to obtain an extremely small amount of the catalyst manganese(IV) oxide (MnO_2), the size of a match head, and place it in a beaker. Place the test tube into the same beaker and measure the mass of the beaker with its contents. Record your observations. Pour the manganese(IV) oxide into the test tube and observe the reaction. After a few seconds, insert a glowing splint into the mouth of the test tube upon which the splint will burst into flames. Place the test tube back into the beaker and measure the mass of the beaker once again. Record your observations.

Reaction D

5. Obtain a piece of magnesium ribbon approximately 1 cm long in a watch glass. Measure the mass of the watch glass with its contents. Record your observations. Using crucible tongs, hold the ribbon in the burner flame until the magnesium starts to burn. DO NOT LOOK DIRECTLY AT THE FLAME. Hold the burning magnesium directly over the watch glass. When the ribbon stops burning, place the remains on the watch glass. Measure the mass of the watch glass once again. Record your observations. Turn off the burner.
6. Clean your test tubes and lab station. Wash liquids down the drain. Wrap solids in a paper towel before discarding them in the trashcan.

Types Of Reactions Lab

K Morrison

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Types Of Reactions Lab:

ChemDiscovery Teacher Edition Olga I. Agapova,2002 **A Concrete Stoichiometry Unit for High School Chemistry** Jennifer Louise Pakkala,2006 **Top Shelf** Brian Pressley,2003 Covers chemical formulas and equations chemical reactions structure of atoms the gas laws and more Presents hands on activities as catalysts to fuel student imagination **The Psychological Index** ,1895 **Oxygen Reduction Reaction** Kushal Sengupta,Sudipta Chatterjee,Kingshuk Dutta,2022-06-14 Oxygen Reduction Reaction Fundamentals Materials and Applications covers the design synthesis and performance efficacies of the entire spectrum of oxygen reduction catalysts extrapolating down to their applications in practical alternative renewable energy devices Catalysts covered include heme inspired iron based heme inspired non iron based non heme based noble metal based non noble metal based and metal free homogeneous and heterogeneous catalysts The book contains critical analyses and opinions from experts around the world making it of interest to scientists engineers industrialists entrepreneurs and students Discusses the fundamental aspects of oxygen reduction reactions Offers a comprehensive analysis of the choice and development of catalyst materials for oxygen reduction reaction Reviews emerging catalyst systems for oxygen reduction reaction Includes analyses of catalytic performance parameters to evaluate their efficacy in oxygen reduction reactions under varied operating conditions Covers the importance of oxygen reduction reaction catalysts and processes in real life applications *Psychological Index* ,1896 **The Psychological Review. Monograph Supplement** ,1896 Psychological Index Howard Crosby Warren,Clement Leslie Vaughan,Knight Dunlap,Madison Bentley,Walker Samuel Hunter,1895 Anatomy & Physiology Laboratory Manual and E-Labs E-Book Kevin T. Patton,2018-01-24 Using an approach that is geared toward developing solid logical habits in dissection and identification the Laboratory Manual for Anatomy and coloring exercises offer a kinesthetic experience useful in retention of content User friendly spiral binding allows for hands free viewing in the lab setting Step by step dissection instructions with accompanying illustrations and photos cover anatomical models and fresh or preserved specimens and provide needed guidance during dissection labs The dissection of tissues organs and entire organisms clarifies anatomical and functional relationships 250 illustrations including common histology slides and depictions of proper procedures accentuate the lab manual s usefulness by providing clear visuals and guidance Easy to evaluate tear out Lab Reports contain checklists drawing exercises and questions that help you demonstrate your understanding of the labs you have participated in They also allow instructors to efficiently check student progress or assign grades Learning objectives presented at the beginning of each exercise offer a straightforward framework for learning Content and concept review questions throughout the manual provide tools for you to reinforce and apply knowledge of anatomy and function Complete lists of materials for each exercise give you and your instructor a thorough checklist for planning and setting up laboratory activities allowing for easy and efficient preparation Modern anatomical imaging techniques such as computed tomography CT magnetic resonance imaging

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Experiment Station Record United States. Office of Experiment Stations, 1911

Toxicology Research Projects Directory ,1979

International Handbook of Research on STEAM Curriculum and Practice Stephen J. Farenga, Salvatore G. Garofalo, Daniel Ness, 2025-10-24 This comprehensive handbook delves into curriculum praxis human development and cognition within the contexts of the STEAM disciplines science technology engineering arts architecture and mathematics Cutting edge research will help educators identify best practice techniques for developing students knowledge in STEAM subjects as well as capture contemporary social and political issues within the STEAM context Drawing on the work of over 50 international contributors this volume covers both emergent and established areas of research giving voice to newcomers to the field as well as perspectives from established experts These areas are divided into five sections on foundations content teaching and learning throughout the lifespan equity and enrichment and settings Each topic is considered in both its historical and current context with a focus on the

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