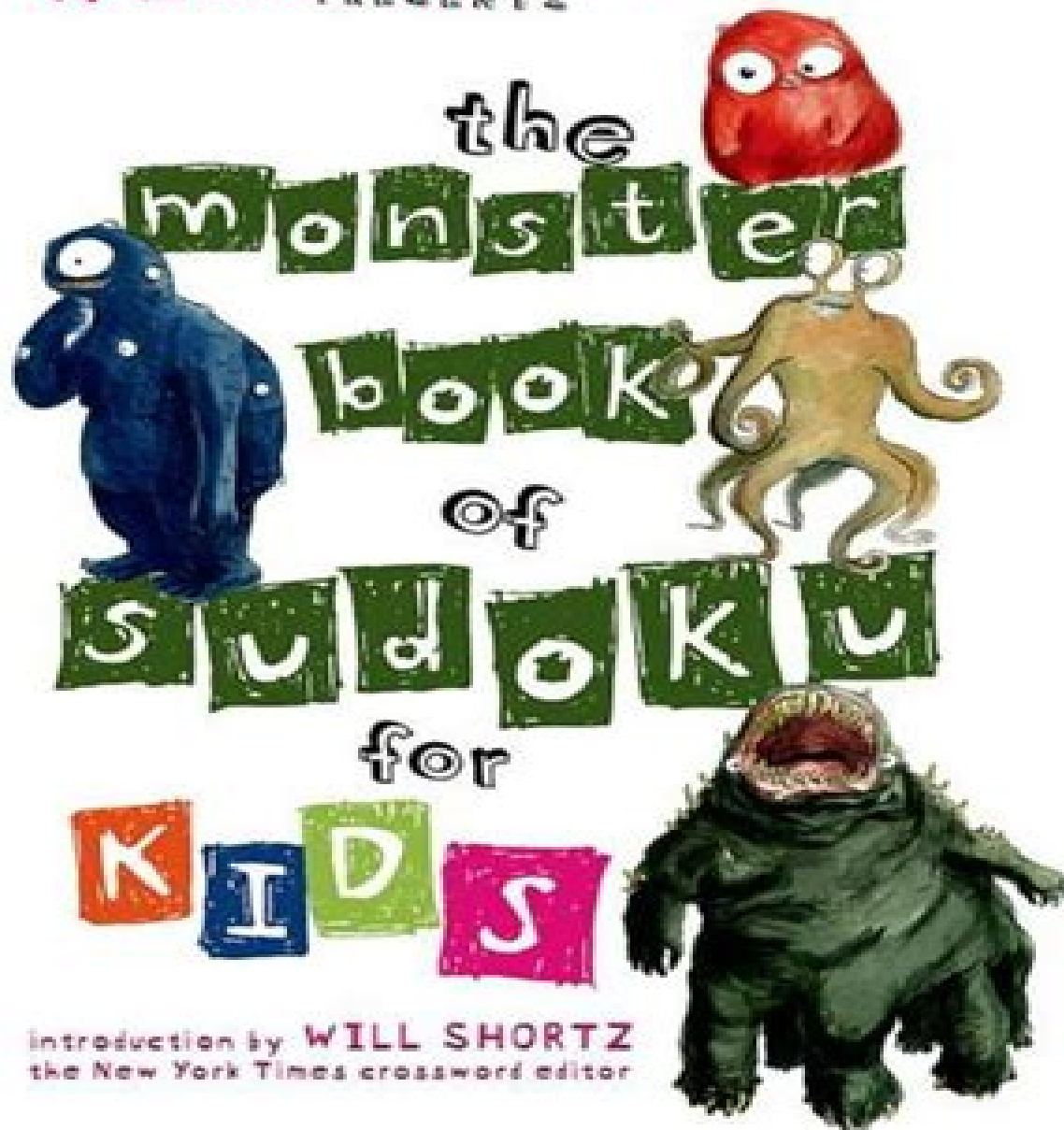


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web so basing on my understanding of this so far an atom is an element that has the same number of protons and neutrons an ion is an element that has different numbers of protons and electrons an ion is positive when it has more protons than electrons and negative when it has more electrons than ions while isotopes are the same element with

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web figure pageindex 2 the formation of a chlorine ion on the left the chlorine atom has 17 electrons on the right the chloride ion has 18 electrons and has a 1 charge with two oppositely charged ions there is an electrostatic attraction between them because opposite charges attract

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