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web linear geometry all angles 180 degrees three regions of electron density around a central atom in a molecule form a trigonal planar geometry all angles 120 degrees four regions of electron density around a central atom in a molecule form a tetrahedral geometry all angles 109 5 degrees

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web in this investigation you will examine a theory that chemists use to explain different aspects of chemical bonding valence shell electron pair repulsion vsepr theory attention will be given to how molecules are arranged in different shapes and how chemists can predict the geometry of a given molecule