#### CHEMISTRY Ch. 13 Notes: Water and Its Solutions

NOTE: Vocabulary terms are in boldface and underlined. Supporting details are in italics.

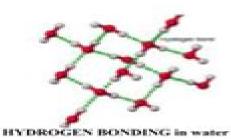
#### 13.1 Notes:

Water Molecule Characteristics П.





POLAR molecule (a dipole)



ICE CRYSTAL

- A) colorless and odorless
- B) neutral pH of 7
- C) triatomic (three atoms)
- D) aregular shape with two unshared electron pairs
- E) polar ( 8+ and 8- areas)
- F) hydrogen bonds—an attraction between hydrogen and an unshared pair of an electronegative element on a neighboring molecule
- G) surface tension—attraction between molecules on the surface of a liquid
  - 1) surface tension makes water bead
  - 2) surfactants-"wetting agents" which decrease surface tension by breaking hydrogen bonds (soaps)
- H) capillarity (capillary action)-moving upward, against gravity (up through roots, etc.)
- hexagonal crystals
- J) high specific heat: 4.184 J/g°C
- K) high boiling point: 100 °C
- HI. Heat Capacity of Water
  - A) heat capacity—amount of heat required to change a substance's temperature by exactly 1 °C

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