

Derive the following equations:

1. Show the following thermodynamic equations:

$$d(\Delta G_{\text{mix}}) = Td(\Delta S_{\text{mix}}) - d(\Delta H_{\text{mix}}) + d(\Delta PV_{\text{mix}}) = -d(\Delta T) \Delta S_{\text{mix}}$$

(a) is this equation a differential or fundamental thermodynamic equation

$$d(\Delta G_{\text{mix}}) = d(\Delta H_{\text{mix}}) - Td(\Delta S_{\text{mix}}) + \Delta S_{\text{mix}} dT$$

(b) what equation the mixed partials of the thermodynamic equations derived?

$$d(\Delta G_{\text{mix}}) = \frac{\partial(\Delta G_{\text{mix}})}{\partial T} dT = -\Delta S_{\text{mix}} dT$$

(c) what equation the mixed partials of the thermodynamic equations derived?

$$\Delta G_{\text{mix}} = \frac{\partial(\Delta G_{\text{mix}})}{\partial T} T = \frac{\partial(\Delta G_{\text{mix}})}{\partial T} T = \Delta S_{\text{mix}} T$$

(d) what is the thermodynamic equation for the mixed partials of the thermodynamic equations?

$$\Delta G_{\text{mix}} = \frac{\partial(\Delta G_{\text{mix}})}{\partial T} T = \frac{\partial(\Delta G_{\text{mix}})}{\partial T} T = \Delta S_{\text{mix}} T$$

2. Thermodynamic equations: $d(\Delta G_{\text{mix}}) = Td(\Delta S_{\text{mix}}) - d(\Delta H_{\text{mix}}) + d(\Delta PV_{\text{mix}}) = -d(\Delta T) \Delta S_{\text{mix}}$

is this a differential or fundamental thermodynamic equation? (which one is the independent variable and which one is the dependent variable)

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a. Show a thermodynamic equation for the mixed partials of the thermodynamic equations.

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