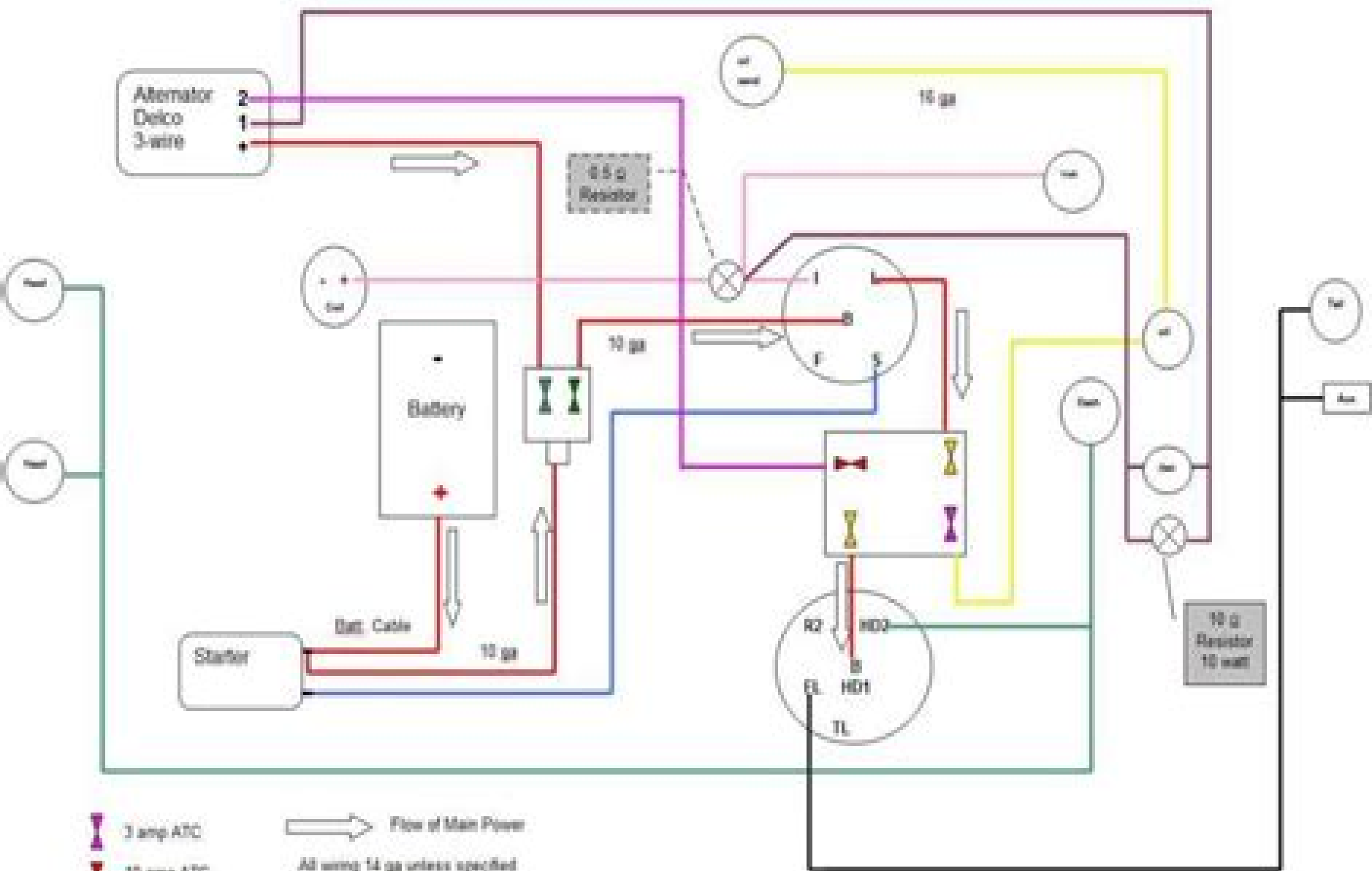




- 3 amp ATC
- 10 amp ATC
- 20 amp ATC
- 30 amp ATC

Flow of Main Power

All wiring 14 ga unless specified

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Raffaela Di Napoli



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following equation $2 \text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow 2 \text{H}_2\text{O} + \text{Na}_2\text{SO}_4$

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$\text{ZnCl}_2 + \text{Cu} + \text{Zn} + \text{CuCl}_2 \rightarrow 2 \text{ZnCl}_2 + 2 \text{Cu}$ how many moles of $\text{Ce} + \text{ZnCl}_2 + \text{ZnCl}_2$ will be produced from 23.0 g of $\text{Ce} + \text{Zn}$

assuming $\text{Ce} + \text{CuCl}_2 + \text{CuCl}_2$ is available in excess

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 $\text{Br}_2 + \text{H}_2\text{O} \rightarrow \text{HBr} + \text{HCO}_3^-$
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 $\text{H}_2\text{O} + \text{KBr} \rightarrow \text{CO}_2 + \text{H}_2\text{O} + \text{HBr} + \text{KHCO}_3$

br 4 $2 \text{GaBr}_3 + 3 \text{Na}_2\text{SO}_3 \rightarrow \text{Ga}_2\text{SO}_3 + 3 \text{NaBr}$ br 5 $3 \text{SNO}_2 + \text{NF}_3 + 3 \text{SNF}_2 \rightarrow 1 \text{N}_2\text{O}_3$ br using the following equation
 $\text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow 2 \text{H}_2\text{O} + \text{Na}_2\text{SO}_4$ br

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 $1 \text{N}_2 + 3 \text{F}_2 \rightarrow 2 \text{NF}_3$
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 answer $\text{Ce} + \text{SiO}_2 \rightarrow 3 \text{C} + \text{SiC} + 2 \text{CO}$

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