

Balancing Equations Worksheet

- 1) $\text{H}_3\text{PO}_4 + \text{KOH} \rightarrow \text{K}_3\text{PO}_4 + \text{H}_2\text{O}$
- 2) $\text{K} + \text{B}_2\text{O}_3 \rightarrow \text{K}_2\text{O} + \text{B}$
- 3) $\text{HCl} + \text{NaOH} \rightarrow \text{NaCl} + \text{H}_2\text{O}$
- 4) $\text{Na} + \text{NaNO}_3 \rightarrow \text{Na}_2\text{O} + \text{N}_2$
- 5) $\text{C} + \text{S}_8 \rightarrow \text{CS}_2$
- 6) $\text{Na} + \text{O}_2 \rightarrow \text{Na}_2\text{O}$
- 7) $\text{N}_2 + \text{O}_2 \rightarrow \text{N}_2\text{O}_5$
- 8) $\text{H}_3\text{PO}_4 + \text{Mg}(\text{OH})_2 \rightarrow \text{Mg}_3(\text{PO}_4)_2 + \text{H}_2\text{O}$
- 9) $\text{NaOH} + \text{H}_2\text{CO}_3 \rightarrow \text{Na}_2\text{CO}_3 + \text{H}_2\text{O}$
- 10) $\text{KOH} + \text{HBr} \rightarrow \text{KBr} + \text{H}_2\text{O}$
- 11) $\text{Na} + \text{O}_2 \rightarrow \text{Na}_2\text{O}$
- 12) $\text{Al}(\text{OH})_3 + \text{H}_2\text{CO}_3 \rightarrow \text{Al}_2(\text{CO}_3)_3 + \text{H}_2\text{O}$
- 13) $\text{Al} + \text{S}_8 \rightarrow \text{Al}_2\text{S}_3$
- 14) $\text{Cs} + \text{N}_2 \rightarrow \text{Cs}_3\text{N}$
- 15) $\text{Mg} + \text{Cl}_2 \rightarrow \text{MgCl}_2$
- 16) $\text{Rb} + \text{RbNO}_3 \rightarrow \text{Rb}_2\text{O} + \text{N}_2$
- 17) $\text{C}_6\text{H}_6 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
- 18) $\text{N}_2 + \text{H}_2 \rightarrow \text{NH}_3$
- 19) $\text{C}_{10}\text{H}_{22} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
- 20) $\text{Al}(\text{OH})_3 + \text{HBr} \rightarrow \text{AlBr}_3 + \text{H}_2\text{O}$
- 21) $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_3 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
- 22) $\text{C}_7\text{H}_8 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
- 23) $\text{Li} + \text{AlCl}_3 \rightarrow \text{LiCl} + \text{Al}$
- 24) $\text{C}_2\text{H}_6 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
- 25) $\text{NH}_4\text{OH} + \text{H}_3\text{PO}_4 \rightarrow (\text{NH}_4)_3\text{PO}_4 + \text{H}_2\text{O}$
- 26) $\text{Rb} + \text{P} \rightarrow \text{Rb}_3\text{P}$
- 27) $\text{CH}_4 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
- 28) $\text{Al}(\text{OH})_3 + \text{H}_2\text{SO}_4 \rightarrow \text{Al}_2(\text{SO}_4)_3 + \text{H}_2\text{O}$
- 29) $\text{Na} + \text{Cl}_2 \rightarrow \text{NaCl}$
- 30) $\text{Rb} + \text{S}_8 \rightarrow \text{Rb}_2\text{S}$
- 31) $\text{H}_3\text{PO}_4 + \text{Ca}(\text{OH})_2 \rightarrow \text{Ca}_3(\text{PO}_4)_2 + \text{H}_2\text{O}$
- 32) $\text{NH}_3 + \text{HCl} \rightarrow \text{NH}_4\text{Cl}$
- 33) $\text{Li} + \text{H}_2\text{O} \rightarrow \text{LiOH} + \text{H}_2$
- 34) $\text{Ca}_3(\text{PO}_4)_2 + \text{SiO}_2 + \text{C} \rightarrow \text{CaSiO}_3 + \text{CO} + \text{P}$
- 35) $\text{NH}_3 + \text{O}_2 \rightarrow \text{N}_2 + \text{H}_2\text{O}$
- 36) $\text{FeS}_2 + \text{O}_2 \rightarrow \text{Fe}_2\text{O}_3 + \text{SO}_2$
- 37) $\text{C} + \text{SO}_2 \rightarrow \text{CS}_2 + \text{CO}$

Unit 7 Balancing Chemical Reactions Worksheet 2

Answers

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web the national certificate vocational for fitting turning comprises four engineering subjects including practical assessments in the fitting turning workshop fitters turners are highly skilled artisans who manufacture and construct components for machinery and use tools such as lathes milling and drilling machines

mechanical engineering n1 n6 studies and course - Feb 05 2023

web the compulsory subjects for fitting and turning n1 n3 are mathematics engineering science and engineering drawing and for n1 n2 you need to do fitting and machining however in n3 fitting and machining is replaced by mechanotechnology

fitting and turning nqf level 3 department of - Mar 06 2023

web these objectives are to create an integrated national framework for learning achievements facilitate access to and progression within education training and career paths enhance the quality of education and training redress unfair discrimination and past imbalances and thereby accelerate employment opportunities contribute to