

**Solution Manual for Chemistry Structure and Properties  
2nd Edition by Tro ISBN 0134293932 9780134293936**

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**Chapter 1. Atoms**

**Student Objectives**

**1.1 A Particulate View of the World: Structure Determines Properties**

- Define **atoms**, **molecules**, and the science of **chemistry**.
- Represent a simple molecule, water, using spheres as atoms.

**1.2 Classifying Matter: A Particulate View**

- Define **matter** and distinguish between the three main states of matter: solid, liquid, and gas.
- Define and understand the difference between **crystalline** and **amorphous** solids.
- Define **mixture**, **pure substance**, **element**, **compound**, **heterogeneous**, and **homogeneous**.
- Differentiate between mixtures and pure substances; elements and compounds; and heterogeneous and homogeneous mixtures.
- Use the scheme on page 000 to classify matter.

**1.3 The Scientific Approach to Knowledge**

- Define and distinguish between a **hypothesis**, a **scientific law**, and a **theory**.
- Understand the role of experiments in testing hypotheses.
- State and understand the law of mass conservation as an example of scientific law.
- Understand that scientific theories are built from strong experimental evidence and that the term "theory" in science is used much differently than in pop culture.
- Understand the importance of reporting correct units with measurements.

# Tro 2nd Edition Solution Manual

**Michael Brown**



## **Tro 2nd Edition Solution Manual:**

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