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web in order to quantify the heat transfer q that is associated with changing the temperature of a substance the mass m and specific heat c of that substance must be multiplied by

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web use this calculator to determine the heat transferred to or from a substance $q = c\Delta t$ in joules j kilojoules kj british thermal units btu iso calories cal or kilocalories kcal

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conduction also known as thermal conduction is the transfer of internal energy within a material the energy is transferred by

collision of molecules atoms and electrons inside the material the heat flow will occur in solid liquid and plasma phases and the energy will always flow from hot to cold joseph see more

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web if the reaction releases heat $q_{rxn} < 0$ then heat is absorbed by the calorimeter $q_{calorimeter} > 0$ and its temperature increases conversely if the reaction absorbs heat

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