

**Energy and Thermal Energy Review**

|                       |                                                                                             |
|-----------------------|---------------------------------------------------------------------------------------------|
| <b>Thermal Energy</b> | 1. the sum of the kinetic and potential energy of the molecules of a substance              |
| <b>Conduction</b>     | 2. transfer of heat by particles colliding with each other                                  |
| <b>Convection</b>     | 3. transfer of heat by the movement of a gas or a liquid                                    |
| <b>Radiation</b>      | 4. transfer of energy by electromagnetic waves                                              |
| <b>Thermometer</b>    | 5. an instrument used to measure temperature                                                |
| <b>Conductor</b>      | 6. a substance that conducts heat                                                           |
| <b>Insulator</b>      | 7. a material through which heat does not easily flow                                       |
| <b>Heat</b>           | 8. thermal energy transferred because of a difference in temperature                        |
| <b>Fahrenheit</b>     | 9. temperature scale with 180 degrees between the freezing and boiling of H <sub>2</sub> O  |
| <b>Celsius</b>        | 10. temperature scale with 100 degrees between the freezing and boiling of H <sub>2</sub> O |
| <b>Temperature</b>    | 11. the average kinetic energy of the particles of a substance                              |
| <b>Specific Heat</b>  | 12. amount of heat to raise 1 kg of a substance 1°C                                         |

13. As matter is heated, its particles move **faster**.

14. Particles that make up matter are in a state of **constant motion**.

15. The energy of moving particles is called **kinetic energy**.

16. The energy stored inside a battery is **chemical** energy.

17. Temperature is a measure of the **average kinetic energy** of atoms and molecules.

18. When a liquid is heated it will (expand or contract)? **expand**

19. A pot on a stove gets hot by means of **conduction**.

20. Since water has a higher specific heat than sand, water will heat **slower** than sand.

21. What are the energy transformations in the follow: a nuclear reaction generates heat to produce steam. The steam is used to turn a generator which finally produces electricity.

**Nuclear to Thermal to Kinetic to Energy**

22. Photosynthesis transforms electromagnetic energy from the Sun into **chemical** energy.

# Unit 3b Thermal Energy Powerpoint Review Worksheet Answers

**Tobias Bleicker**



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